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Hot workability and dynamic recrystallization mechanisms of N6

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Abstract

The hot workability and dynamic recrystallization (DRX) mechanisms of N6 were systematically investigated using thermal compression tests. Based on hot compression data, the N6 constitutive equation was developed and its reliability was verified. As its processing map was constructed and combined with microstructural observations a semi-quantitative response relationship between thermal deformation parameters and microstructure was established. The DRX process of N6 is a thermally activated process and particularly sensitive to the strain rate. The optimal thermal processing parameters for N6 were established as 950-1050 ℃/0.1~1 s^{-1}. Furthermore, it was proven that the dominant nucleation mechanism is discontinuous dynamic recrystallization (DDRX) characterized by grain boundary bulging and twins assisting nucleation, while the continuous dynamic recrystallization (CDRX) characterized by subgrains combined with rotation is an inactive nucleation mechanism.

Key words: Microstructure evolution, Hot workability, Processing map, Dynamic recrystallization

1. Introduction

N6 is widely used in the alkali industry due to its incomparably superior corrosion resistance to high temperature and high concentrations of caustic alkali. Its large thick-walled pipes are mostly used in evaporators, heaters, lye storage tanks, pipelines, pumps, and valves in the alkali industry [1-3]. However, current research focuses on nickel-based alloys. The mechanical properties of N6 are lower than other nickel-based alloys, which limits its application in practical engineering [4, 5]. Therefore, the lack of systematic and in-depth research on N6 leads to unclear mechanisms of microstructure transformation of its tubes,

which seriously restricts the performance improvement of relevant equipment and the demand for high-quality pure nickel tubes and bars. The hot forming process is an optimal method to manufacture N6 tubes or tube components. Similar to many nickel-based superalloys, the thermal deformation of N6 is characterized by complex dynamic recrystallization (DRX) structural evolution, which is sensitive to deformation temperature (T), strain rate ($\dot{\varepsilon}$), and strain (ε) [6-10]. Therefore, reasonable process parameters are necessary to achieve high-quality manufacturing of N6 pipes or pipe components.

The constitutive model and processing map based on the dynamic material model (DMM) are key components in understanding the thermal deformation behavior of the material [11-15]. In recent years, many scholars have used the classic Arrhenius hyperbolic sine formula and processing maps to predict the high temperature deformation behavior of various nickel-based alloys, such as GH4169 alloy [11], 80A alloy [12], GH4698 alloy [13], GH696 alloy [14], and Inconel 625 [15], etc.

DRX is the main softening mechanism of nickel-based alloys with lower stacking fault energy (SFE). In previous studies, conventional discontinuous dynamic recrystallization (DDRX) characterized the local bulge of grain boundaries as the main nucleation mechanism in nickel-based alloys [16-19]. In addition, twins can also promote the nucleation of nickel-based alloys by accelerating the protrusion of the original grain boundary and the subsequent separation of the protrusion from other grain boundaries [20,21]. On the other hand, continuous dynamic recrystallization (CDRX), characterized by progressive rotation of subgrains, has also been shown in other nickel-based superalloys over the past few years. Although CDRX is considered to occur easily in high SFE metal materials [17, 22-26], there is no consensus on the favorable deformation conditions of CDRX in nickel-based superalloys. Therefore, an in-depth understanding of the hot forming behavior and DRX mechanism in the N6 hot deformation process is urgently needed.

In this study, the thermal deformation behavior and DRX mechanism of N6 alloys were studied through a series of isothermal compression experiments. Based on a constitutive model and thermal processing map, the thermal deformation behavior of N6 was studied. The focus was on the DRX mechanism in N6 thermal deformation, and the typical microstructures involved in different DRX mechanisms are discussed in depth. This research provides

theoretical guidance for the determination of thermal deformation process parameters and organization control for the preparation of N6 pipes.

2. Materials and methods

The nominal chemical compositions (wt.%) of N6 are 0.047C-0.05Mn-0.018Si-0.064Cu-0.3Co-0.046Al-0.11Fe-0.018Ti-0.045S and (Bal.) Ni. The specimens of Ø8×12 mm were taken from a hot rolled billet. Figure 1(a) is the typical metallographic microstructure of N6, which consists of nearly equiaxed, static recrystallization (SRX) grains (yellow) and a multitude of annealing twins (Fig. 1(b)). The hot deformation tests were performed in a Gleeble-3500 thermal simulator, which were carried out over a temperature range of 900 ℃-1200 ℃ (in increments of 100 ℃) and strain rates of 0.01-10 s^{-1} with a true strain of 0.7. The hot compression process scheme is shown in Fig. 1(b).

After thermal compression, these specimens utilized the EDM wire cutting equipment to cut along the compression axis for microstructural analysis. The observation location was at the central region of the section mentioned above. The metallographic specimens were ground using various grits of sandpaper (400# to 3000#). Afterwards, the specimens were polished using a 0.5 μm diamond solution and etched in a solution of 3 mL HNO_3 + 5 mL $C_2H_4O_2$ for 30 s, and then observed using a LSM 800 optical light microscope. For EBSD analysis, samples were electropolished in a solution of 90 mL C_2H_5OH and 10 mL $HClO_4$ at −20 ℃ at a voltage of 50 V. The electron backscatter diffraction (EBSD) measurements and analyses were completed using the HKL system mounted in a FEI NovaNano SEM 430. The sample microstructure was characterized on the compression direction-transverse direction (CD-TD) cross section. The microstructure of the sample at 900 ℃/1s^{-1} was observed by TEM (JEM-2100) with a working voltage of 200 kV. The TEM sample was first sliced to a thickness of about 50 μm by sandpaper and then punched into 3 mm diameter discs. This was followed by double-jet electrolytic polishing at 0 ℃ using an electrolyte consisting of 90 mL C_2H_5OH and 10 mL $HClO_4$.

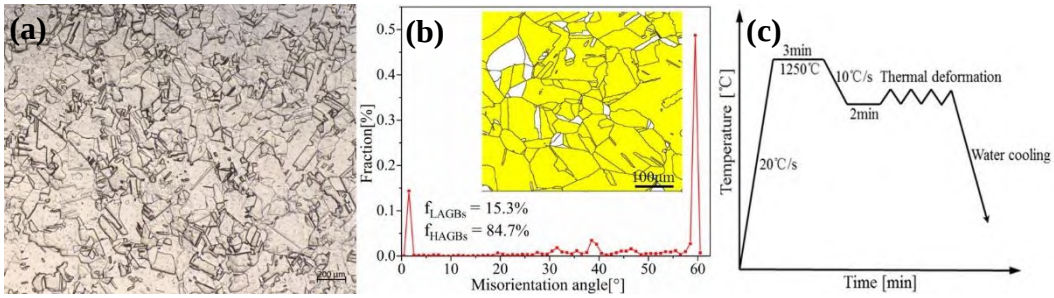

Fig. 1 (a) The metallographic microstructure of N6; (b) the microstructure of N6 characterized by EBSD and (c) the hot compression process scheme

3. Results

3.1 Hot deformation behavior

Fig. 2 shows the true stress-strain curves of the N6 hot compression test. It can be seen that the stress level increases rapidly to a peak value with increasing strain, then decreases, and finally reaches steady state. This trend corresponds to three distinct stages: work hardening, softening process, and stability [6, 15, 27]. In the initial stage of deformation, work hardening plays a primary role, which causes dislocations to increase and rapidly accumulate, leading to a rapid increase in flow stress. DRX occurs when the dislocations accumulate to a value [28]. The action of DRX gradually reduces the flow stress. Under the combined action of work intensification and DRX, the flow stress reaches a stable state. The rheological curve of N6 shows obvious DRX characteristics.

At a high strain rate (10 s^{-1}), the flow stress shows a tendency to increase and then decrease, and it is difficult to reach a steady-state rheological state. This may be due to the short deformation time of high strain rate (10 s^{-1}), the instantaneous change of stress and strain, and the generation of deformation heat, but the deformation heat does not have enough time to conduct outwards, resulting in a significant temperature increase in some areas. This phenomenon is called the adiabatic temperature rise (ATR) effect [29]. The heat of deformation can be estimated using the following formula:

$$\Delta T = \frac{\eta \int \sigma d\varepsilon}{\rho C_p} \tag{1}$$

where ΔT is the temperature field, $\int \sigma d\varepsilon$ is the work of plastic deformation, ρ is the density of the material, C_p is the specific heat capacity of the material, and η is the adiabatic factor, generally 0.9. The adiabatic temperature rise as a function of different temperature thermal deformation at strain rates of 10s^{-1} is shown in Figure 3 (a).

The ATR phenomenon leads to changes in the measured flow stress, which should be corrected for temperature rise [30]. The influence of temperature rise on stress is expressed by equation 2:

$$\Delta\sigma = \frac{Q(\frac{1}{T}-\frac{1}{T+\Delta T})}{Rn\alpha} \tag{2}$$

where $\Delta\sigma$ is the stress difference caused by adiabatic temperature rise, MPa; Q is the activation energy of thermal deformation, kJ·mol^{-1}; R is the gas constant, with a value of 8.314 J·mol^{-1}·K^{-1}; T is the deformation temperature, K; and n and α are used to describe the rheological properties of the material when it is deformed at high temperature.

According to equation (2), the corresponding stress value at the actual temperature can be obtained, as shown in Figure 3(b). To reduce the number of calculations, only the experimental points whose true strain value is an integer multiple of 0.05 are used to correct the stress and temperature rise.

Using metals as an example, the Johnson–Cook (JC) model has become a widely used phenomeno-logical model to describe a material's stress–strain response with both strain-rate and temperature effects [31]. The JC model describes how much the material is softened with respect to temperature but does not directly indicate how much the temperature increases. This results in a relatively implicit relationship between strain/strain rate and the thermosoftening effect [32]. Although efforts have been made to solve the thermal softening effect in a more explicit way using the JC model, the results of the research are inconsistent. Many authors relate the adiabatic rise in temperature with a low fraction of twinning at high strain rate compression [33,34]. Other authors mentioned that there is a strong positive correlation between the ATR value and the strain rate, but there is a strong negative correlation between the ATR value and deformation temperature [35]. Based on the experimental foundation and experience of the predecessors, this research has carried out corresponding calculations and corrections on the influence of the adiabatic temperature rise effect.

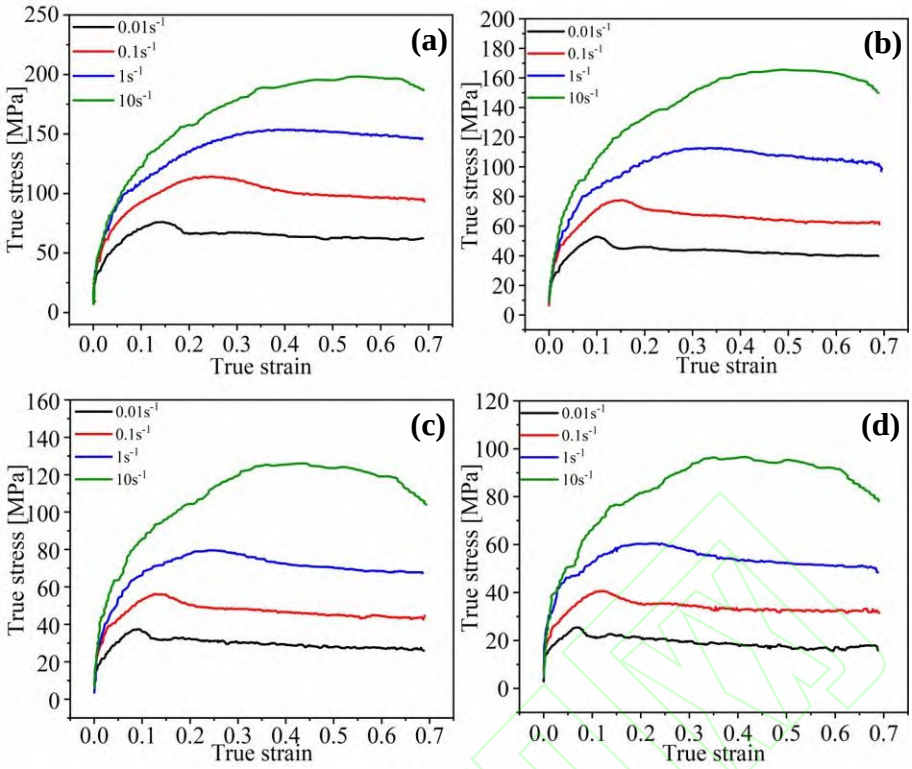

Fig.2 True stress-true strain curves of N6: (a) 900 ℃, (b) 1000 ℃, (c) 1100 ℃, and (d) 1200 ℃

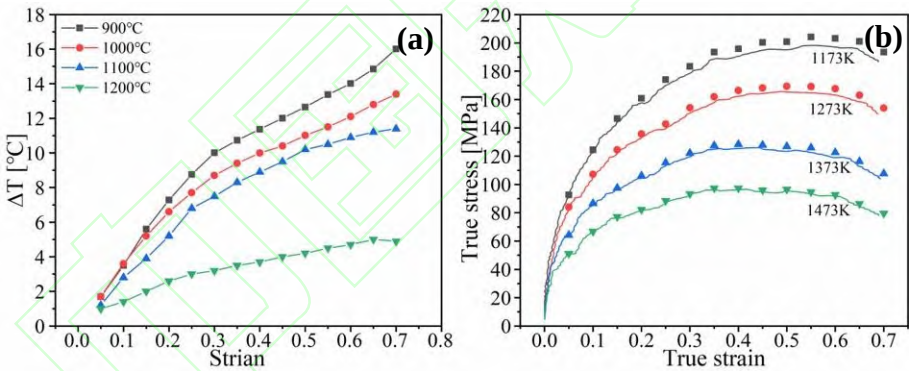

Fig. 3 (a) Adiabatic temperature rise at different temperatures and a strain rate of 10 s^{-1};(b) Adiabatic temperature rise correction stress curve

3.2 Constitutive modeling of hot deformation

The constitutive equation expresses the relationship between flow stress, deformation temperature, and strain rate under hot working conditions, which is the dynamic response to the material deformation process [31]. This article uses the Arrhenius hyperbolic sine function model to describe the general form of the function. These equations are shown as 3-5 below [29]:

$$\dot{\varepsilon} = A_1 \sigma^{n_1} \exp(-\frac{Q}{RT}) \qquad \alpha\sigma<0.8 \qquad (3)$$

$$\dot{\varepsilon} = A_2 \exp(\beta\sigma)\exp(-\frac{Q}{RT}) \qquad \alpha\sigma>1.2 \tag{4}$$

$$\dot{\varepsilon} = A[sinh(\alpha\sigma)]^n \exp(-\frac{Q}{RT}) \qquad \text{for all } \sigma \tag{5}$$

where $\dot{\varepsilon}$ is the strain rate, s^{-1}; A is structure factor, s^{-1}; α is the stress level parameter, MPa^{-1}; σ is the stress, MPa; n is the stress index; Q is the thermal deformation activation energy, $kJ\cdot mol^{-1}$; R is the gas constant, with a value of 8.314 $J\cdot(mol\cdot K)^{-1}$; T is the deformation temperature, K. A_1, A_2, n_1 and β are all constants after conversion, and $\alpha=\beta/n_1$.

Using peek stress σ_p as the input data, the unknown constant parameter in the equation is obtained by the linear regression method. Taking the natural logarithms for both sides of Eqs. (3) - (5), the following are obtained:

$$n_1 = \left[\frac{\partial \ln\dot{\varepsilon}}{\partial \ln\sigma_P}\right]_{\mathrm{T=constant}} \tag{6}$$

$$\beta = \left[\frac{\partial \ln\dot{\varepsilon}}{\partial \sigma_P}\right]_{\mathrm{T=constant}} \tag{7}$$

$$\mathrm{n} = \left[\frac{\partial \ln\dot{\varepsilon}}{\partial \ln[\sin h(\alpha\sigma)]}\right]_{\mathrm{T=constant}} \tag{8}$$

Figure 4 shows the peak stress of N6 at different temperatures and strain rates during high-temperature compression. As seen from the above formula, the values of n_1 and α can be obtained through the relationship between $\ln\sigma$ - $\ln\dot{\varepsilon}$ and σ - $\ln\dot{\varepsilon}$. Linear regression analysis is performed on equations (6) and (7) according to the peak stress measured by the thermal compression experiment, and the linear relationship diagrams of $\ln\sigma$ - $\ln\dot{\varepsilon}$ and σ - $\ln\dot{\varepsilon}$ are fitted, respectively, as shown in Fig. 5(a) and (b). The values of n_1 and β are equal to the reciprocal of the mean of each slope in the graph. Therefore, n_1=6.07632 and β=0.07414 MPa^{-1}. According to $\alpha=\beta/n_1$, α=0.012 MPa^{-1} can be obtained, the value of n can be determined from the relationship of $\ln[\sin h(\alpha\sigma)]$-$\ln\dot{\varepsilon}$ (Fig. 5(c)), and the average value of n is equal to the reciprocal of the mean of each slope in the graph. Therefore, n=4.355.

Taking the natural logarithms of both sides of Eq. (5), the following equation can be obtained:

$$\ln\dot{\varepsilon} = \ln A + n\ln[\sinh(\alpha\sigma)] - Q/RT \tag{9}$$

According to Eq. (9), it can be found that the intercept of the fitted straight line $\ln[\sinh(\alpha\sigma)]$ - $\ln\dot{\varepsilon}$ is lnA, so that the value of structure factor A can be obtained. From Fig. 5(c), the average intercept of the four fitted straight lines can be calculated to be 22.907, that is, lnA=22.907, and then A= $8.879\times10^9 s^{-1}$.

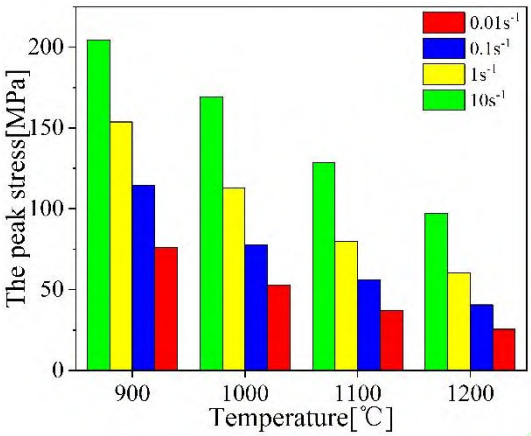

Fig. 4 Peak stress during N6 thermal compression

Taking the partial differential equation of both sides of Eq. (9), the activation energy of hot deformation can be obtained from the following equation:

$$Q = \mathrm{R}\left[\frac{\partial \ln \dot{\varepsilon}}{\partial \ln[\sin h(\alpha\sigma)]}\Big|_{T=\mathrm{constant}}\right]\left[\frac{\partial \ln[sin\, h(\alpha\sigma)]}{\partial T^{-1}}\Big|_{\dot{\varepsilon}=\mathrm{constant}}\right] \quad (10)$$

where R is the gas constant, with a value of 8.314 J·(mol·K)$^{-1}$, $\left[\frac{\partial \ln \dot{\varepsilon}}{\partial \ln[\sin h(\alpha\sigma)]}\Big|_{T=\mathrm{constant}}\right] = n$ =4.355. Similarly, the value of Q can be determined fromthe relationship of $\ln[\sin h(\alpha\sigma)]$ - 1000/T (Fig. 5(d)). According to Eq. (10), the activation energy of deformation Q = 272.988 kJ·mol^{-1}.

Compared with other nickel-based superalloys, pure nickel N6 has a lower thermal deformation activation energy. For example, the thermal deformation activation energy of Ni80A alloy is 405.57 kJ/mol [36]. When the strain is 0.20-0.60, the apparent activation energy of deformation of GH696 superalloy is in the range of 389.524-410.236 kJ/mol [37]. The average activation energy for the thermal deformation of Inconel 740 alloy at 1000 -1200 ℃ is 373.86 kJ/mol [38]. This is mainly due to the addition of other elements in the alloy, resulting in a dislocation movement and thermodynamic mechanism different from that of N6. When the elements of the alloy are determined, the thermal deformation activation energy is mainly related to the deformation temperature and strain rate, so the calculated Q value above is the average value of N6 in the range of 900 -1200 ℃.

According to the above derivation, the constitutive equation in the process of N6 hot deformation can be obtained as:

$$\dot{\varepsilon} = 8.879 \times 10^{9}[sinh(0.0122\sigma)]^{4.355}\exp(-\frac{272.988}{RT}) \quad (11)$$

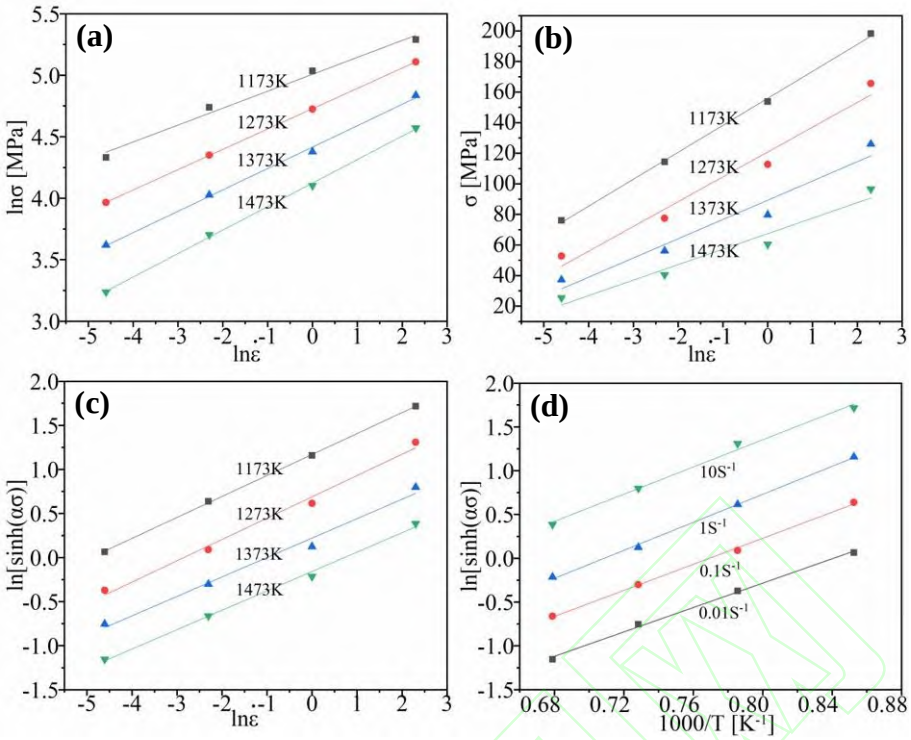

Fig. 5 Linear fitting relationships: (a) $\ln\sigma$ - $\ln\dot{\varepsilon}$, (b)σ - $\ln\dot{\varepsilon}$, (c) $\ln[\sinh(\alpha\sigma)]$ - $\ln\dot{\varepsilon}$, (d) $\ln[\sin h(\alpha\sigma)]$ - $1000/T$

3.3 Thermal processing map

To assess the hot workability of the tested N6, a hot deformation processing map was considered based on the dynamic materials model (DMM) [39, 40]. Under the conditions of constant temperature and strain rate, the constitutive relationship of the material in the process of hot deformation can be expressed as Eq. (12):

$$\sigma = K\dot{\varepsilon}^{m} \qquad (12)$$

where K is a constant and m is a strain rate sensitive factor, which can expressed as the distribution coefficient between the energy consumed by plastic deformation and the energy consumed by tissue changes. m can be expressed as the following relationship [41]:

$$m = \left(\frac{\partial J}{\partial G}\right)_{\varepsilon,T} = \left[\frac{\partial(\ln\sigma)}{\partial(\ln\dot{\varepsilon})}\right]_{\varepsilon,T} = \left(\frac{\dot{\varepsilon}\partial\sigma}{\sigma\partial\dot{\varepsilon}}\right)_{\varepsilon,T} \qquad (13)$$

The power dissipated for microstructural evolution is reflected by power dissipation efficiency η, which is a function of strain rate sensitivity [41]:

$$\eta = \frac{2m}{m+1} \qquad (14)$$

The stress value at the true strain of 0.5 is extracted from the thermal compression data to a series of derivation calculations, where the power dissipation map can be drawn on the

plane formed by the deformation temperature T and the strain rate $\dot{\varepsilon}$ (Fig. 6 (a)).

Only relying on the power dissipation map is not enough to accurately and effectively judge the processing performance of the corresponding parameter area. It is also necessary to use the instability interval for further determination. The criterion of the processing instability zone is given, which can be expressed as the following relationship:

$$\xi(\dot{\varepsilon}) = \frac{\partial \ln\left(\frac{m}{m+1}\right)}{\partial \ln \dot{\varepsilon}} + m =< 0 \quad (15)$$

where ξ is a function of T and $\dot{\varepsilon}$. Similarly, the isoplethic curves of ξ can be drawn on the plane formed by the deformation temperature T and the strain rate $\dot{\varepsilon}$ (Fig. 6(b)). The negative $\xi(\dot{\varepsilon})$ value region is flow instability.

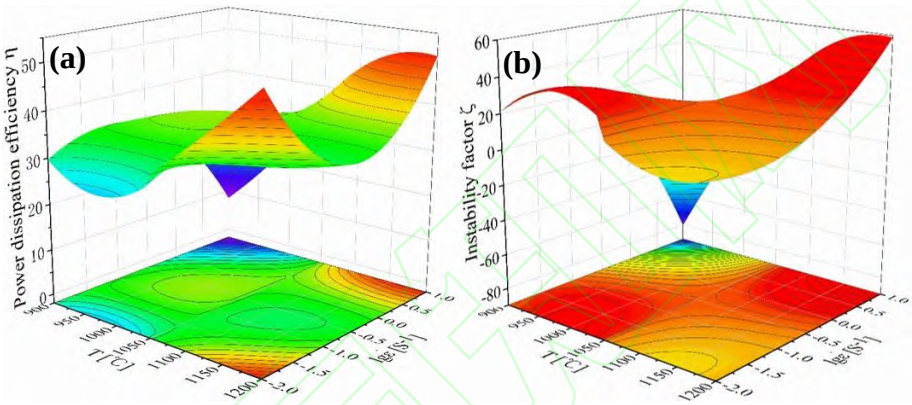

Fig. 6 (a) Power dissipation map of N6 at 0.5 true strain; (b) Instability map of N6 at 0.5 true strain

The processing map can be constructed by the superposition of the instability on the power dissipation map, which can be used for processing parameter optimization and microstructure control in the hot working process. Fig. 7(a) is the processing map of N6 at 0.5 true strain. The shaded part is the instability area, where the darker the color, the greater the degree of instability. The instability area refers to the region where the instability factor ξ is less than 0 in the instability diagram, which is the shaded part in the hot working diagram. One main instability region exists in the processing map, where the temperature range of the instability region is 900~1050 °C, and the strain rate is 1~10 s^{-1}. The metallographic structure corresponding to the instability zone is shown in Fig. 7(b), which can be seen that larger elongated deformed grains (marked by the circle) are surrounded by fine dynamic recrystallized grains. The grain size in this region is not uniform, which may be the primary reason for the processing instability. The region where the value of η decreases sharply in the

processing map is called the processing danger regions. The metallographic structure of processing dangerous region I and II are shown in Fig. 7(d) and (e), respectively, which can be seen as severe grain coarsening, thereby reducing the processing performance of N6. The metallographic structure of the safe region is basically equiaxed, and the grain size is small and uniform (Fig. 7(b)). The range of the safe region is 900~1050 ℃, and the strain rate is 0.1~1 s^{-1}. In summary, the establishment of N6 thermal processing provides theoretical guidance to formulate reasonable thermal processing parameters.

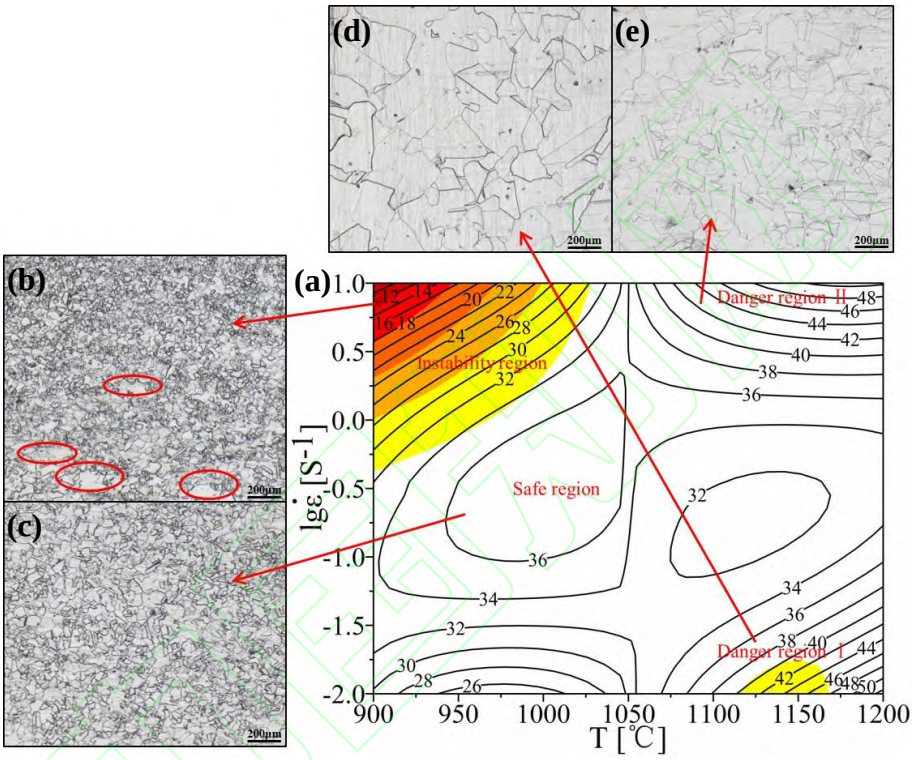

Fig. 7 Thermal processing map and corresponding metallographic structure in different regions: (a) thermal processing map; (b) 900 ℃/1 s^{-1} in the instability region; (c) 1000 ℃/1 s^{-1} in the safe region; (d) 1200 ℃/0.01 s^{-1} in danger region Ⅰ; (e) 1200 ℃/10 s^{-1} in danger region Ⅱ

3.4 Microstructural and microtextural evolution

To study microstructural and microtextural evolution, EBSD observations were carried out on samples at strain rates of 0.01-10 s^{-1} at 900℃, respectively. Fig. 8 shows the orientation imaging microscopy (OIM) maps ((a)-(d)) and orientation angle distribution maps ((e)-(h)) at strain rates of 0.01-10 s^{-1} at 900℃, respectively. The inset of orientation angle distribution maps is the grain boundary map, in which different colors represent different

grain boundaries: the black line represents HAGBs, the red line is Σ3 twin grain boundaries (TBs), and the green line is sub-boundaries (SBs). As shown in Fig. 8(a) and (e), relatively high density SBs (<2°, indicated by green boundaries) are observed in large, deformed grains, which are surrounded by fine DRX grains. At the same time, some of the recrystallized grains have grown due to the lower strain rate. When the strain rates are 0.1 and 1 s^{-1} (Fig. 8(b) and (c)), it can be seen that the deformed grains are slightly elongated in the direction of the compression axis and some smaller equiaxed DRX grains are attached to the jagged grain boundaries of the deformed grains. The dynamic recrystallization structure is described as a "necklace structure," which confirms the occurrence of the DDRX mechanism during the hot working process [42,43]. When the strain rate is 10 s^{-1} (Fig. 8(d)), the elongated deformed grains have been replaced by DRX grains and its grain size is significantly larger than 0.1 s^{-1} and 1 s^{-1} conditions. From Fig. 8(e), (f), (g) and (h), it can be seen that the orientation angle distributions of hot-deformed N6 at different strain rates at 900 ℃ are basically bimodal, but the proportion of different types of GBs greatly varies. In addition, the distribution of Σ3 TBs are in the grain and near grain boundaries. Furthermore, the proportion of LAGBs decrease but HAGBs increase in the orientation angle distribution as the strain rate increases. From a microstructural point of view, the misorientation angle distribution is associated with the kinetics of dynamic recrystallization [44]. Increases in θ_{Avg} correspond to increases in volume fraction of DRX grains and vice versa. Specifically, θ_{Avg} increases from 26.956° (900 ℃, 0.01 s^{-1}) to 40.921° (900 ℃, 10 s^{-1}). The orientation angle distribution and grain boundary maps of N6 with different strain rates at 900℃ clearly show the evolution of the deformed grains are composed of entangled dislocations, subgrains, and LAGBs to the DRX grains.

Fig. 9 shows that the quantitative description of the microtextural evolution is presented by the sections (φ_2 = 0°, 45°, and 65°) of ODFs at 900℃ and different strain rates. Since DRX occurs during thermal deformation, the grains are characterized by relatively randomized orientation. Some ideal texture components are found during the thermal deformation of N6, including shear {011}<110>, Cu {112}<111>, Goss {011}<001>, and Brass {011}<211> [45].

The evolution characteristics of DRX volume fraction and the maximum intensity of all microtextural components at 900 ℃ and different strain rates are shown in Fig. 10. In the

strain rate range of 0.01-10 s^{-1}, it can be found that the volume fraction of DRX grains increases with increasing strain rate. The acceleration of DRX at high strain rate was also observed in 800H alloy [46] and GH690 superalloy [47]. The maximum intensity of all microtextures displays a decrease first and then increased tendency with increasing DRX fraction, which implies the level of DRX fraction plays an obvious role in strengthening the maximum intensities at the high strain rate of all the microtextural components in N6.

4. Discussion

Additional research must be performed to illustrate the specific DRX mechanism in N6 thermal deformation. Furthermore, an in-depth understanding of the specific manifestation of different DRX mechanisms during the hot deformation of N6 is also imperatively needed.

4.1 Discontinuous dynamic recrystallization

4.1.1 Grain boundary bulging nucleation

DDRX is the main nucleation mechanism of nickel-based alloys, where bulging of the original HAGBs and the necklace structure consisting of DRX grains along the original HAGBs are its typical microscopic characteristics [16-18]. Fig. 11 shows a partial IPF map of N6 under 900 °C/0.1 s^{-1} thermal deformation conditions. The typical necklace structure (marked by white ellipse) and strain-induced SBs (marked by white circle) can be observed in the microstructure of Fig. 11(a). Fig. 11(b) is an enlarged view of the white elliptical markers, and it can be observed from the microstructure shown in Fig. 11(b) that the initial HAGBs bulging position can be used as the actual nucleation position of DRX grains. To further characterize the orientation of DRX grains and parent grains, Grain 1 and Region 1 represent DRX grains and adjacent parent regions, respectively. The {111} pole figures in Fig. 11(c) show the orientation relationships between Grain 1 and Region 1, where CD and TD represent the compression direction and transverse direction, respectively. Grain 1 and Region 1 are rotated 17.3° around the [-2 -1 6] axis. When the HAGBs bulge due to different dislocation densities, the bulged region will inherit the orientation relationship of the parent grains [42]. The different rotational degree of the bulging areas result in the transformation of the strain-induced sub-boundary between Grain 1 and Region 1 into the HAGBs. Thereby, DRX grains form.

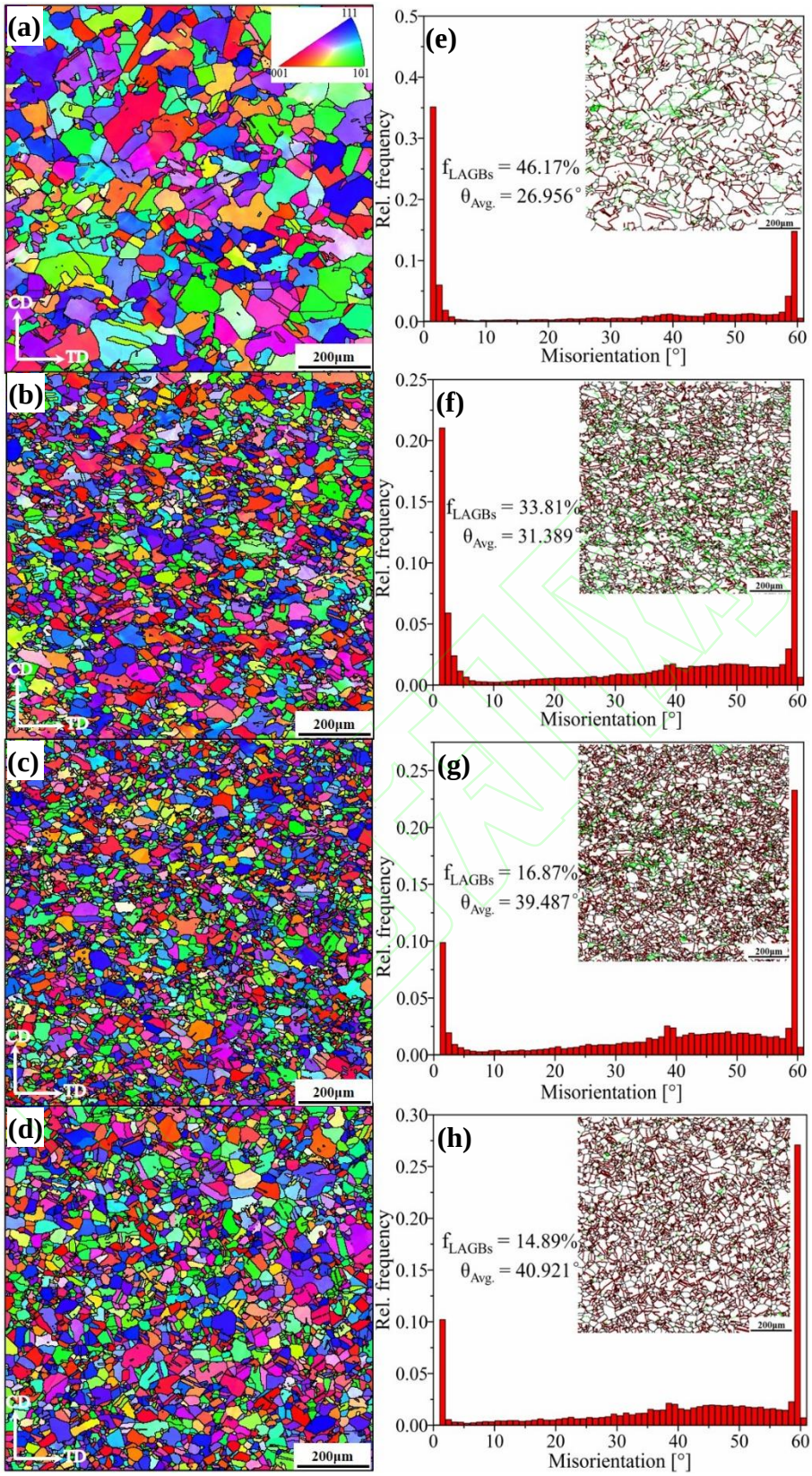

Fig. 8 OIM maps and grain misorientation angle distributions of N6 deformed at 900 °C with the strain rate of 0.01 s^{-1} (a, e), 0.1 s^{-1} (b, f), 1 s^{-1} (c, g), and 10 s^{-1} (d, h)

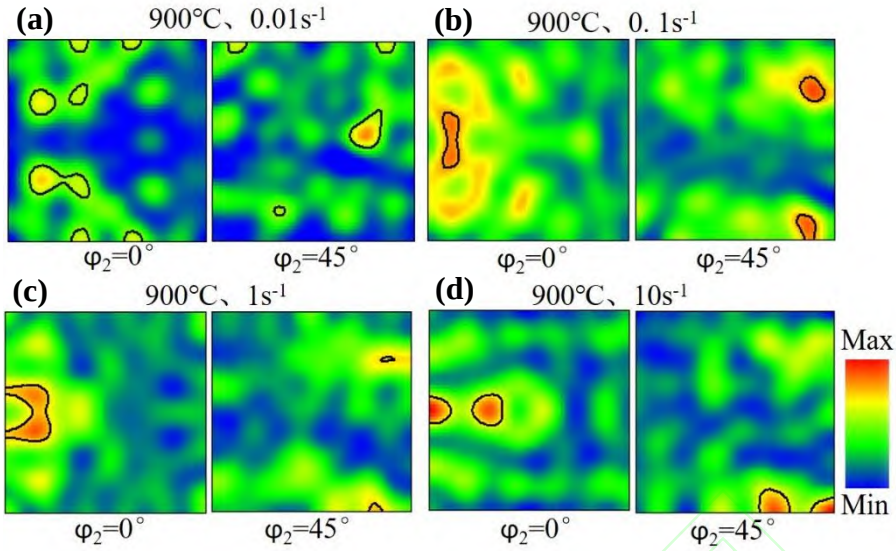

Fig. 9 The ODF represented inφ 2 = 0°, 45°, and 65° of the Euler space under the deformation conditions of different strain rates and 900 ℃: (a) 0.01 s^{-1}, (b) 0.1 s^{-1}, (c) 1 s^{-1} and (d) 10 s^{-1}

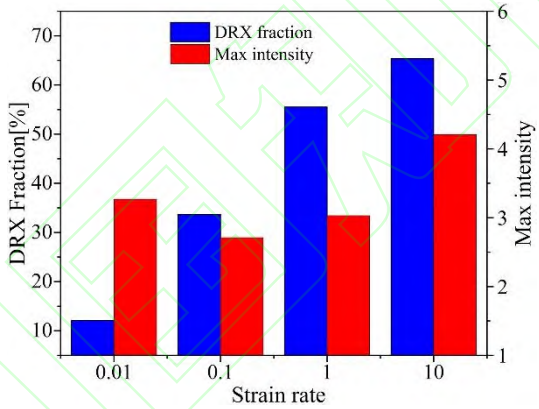

Fig. 10 The relationship between the DRX volume fraction and the maximum intensity of all microtextural components at 900 ℃ -and different strain rates

4.1.2 Twin-assisted nucleation

Fig. 12(a) and (b) reveal that most of the newly formed TBs are located inside the DRX grains, which indicates twinning occurs during the growth stage of DRX grains. Twinning can reduce the boundary energy of growing grains and improve the mobility of grain boundaries [48,49], and the initial TBs will transform into HAGBs due to thermal deformation. As shown in Fig. 12(b), the DRX grain with Σ3 boundary around the deformed grain, which are marked as Grain 1, and the grain adjacent to this Σ3 boundary is marked as Grain 2. It can be seen from Fig. 12(d) that the orientation relationship of two grains is shown in the pole figure of {111},

and the overlapping point indicates the twin relationship between Grain 1 and Grain 2 on the {111} pole plane. Also, Grain 1 and Region 2 are rotated 58.8° around the [-1 1 -1] axis. Moreover, the misorientation angle of the coherent TBs is generally less than 1.7° away from ideal 60° /<111>, i.e., about one-fifth of the Brandon criterion for Σ3 (8.7°) [50]. At the same time, the deformed grains with SBs are attached to the GBs of Grain 2. As shown in Fig. 12(b), the Σ3 GBs in the DRX grains show the smallest distortion, while the deformed grains show a higher density of SBs. A line scan corresponding to the black arrow was conducted on the deformed grains and the measured misorientation profile is depicted in Fig. 12(c). The point to origin profile accumulates misorientations constantly, showing the high misorientation gradient. The difference in internal distortions of the two kinds of grains is the main driving force for grain boundary bulging, which allows the HAGBs to migrate. Therefore, twinning accelerates the DRX process.

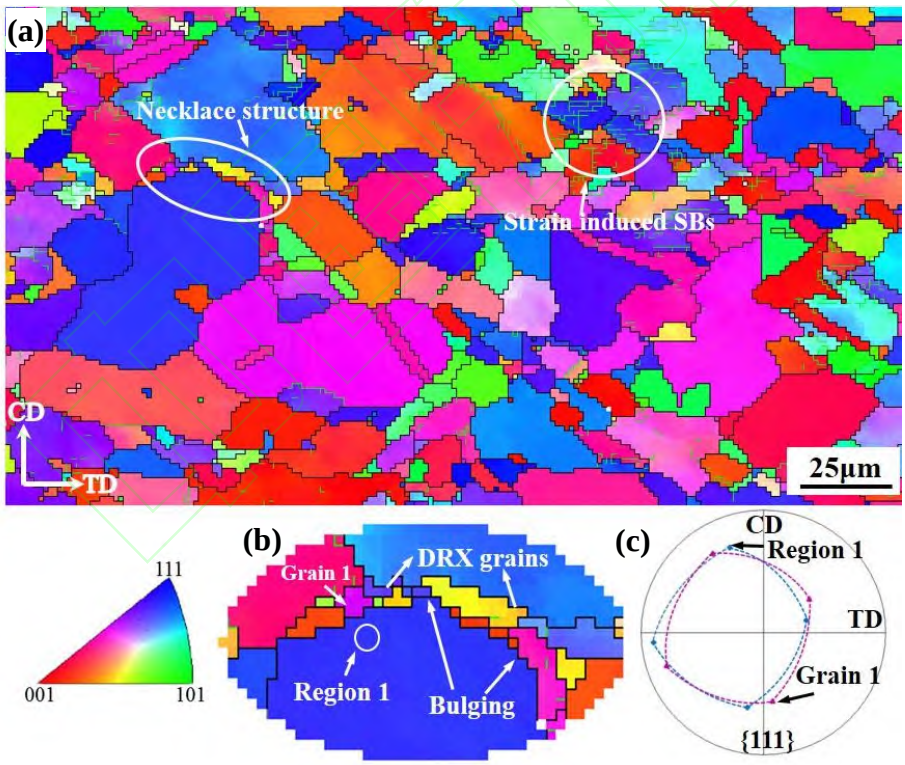

Fig. 11 (a) Microstructure of N6 at 900 °C/0.1 s^{-1}, (b) the enlarged view of the necklace structure, (c) the crystallographic relationship, and (d) between Grain 1 and Region 1

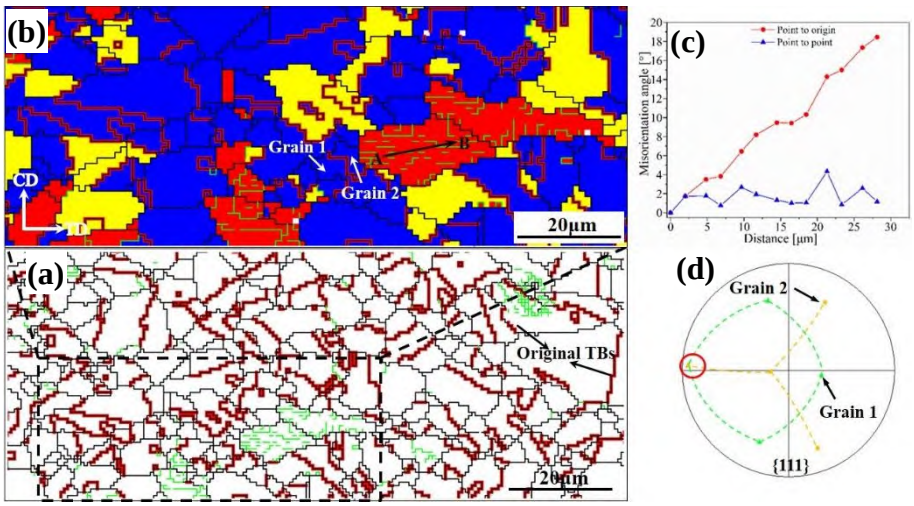

Fig. 12 (a) Grain boundary map of N6 obtained at 900 °C/0.1s^{-1}, (b) the enlarged view of (a) marked by a black rectangle, (c) the misorientation gradient measured along the A-B line, (d) the crystallographic relationship between Grain 1 and Region 1

4.2 Continuous dynamic recrystallization

Studies have shown that DDRX and CDRX often occur simultaneously in the process of thermal deformation of nickel-based superalloys [51]. Fig. 13(b) is a partial enlarged view of Region 1, and it can be seen that DRX grains with HAGBs formed inside the initial deformed grains, which belongs to the typical nucleation mechanism that DRX grains were formed by progressive subgrain rotation. At the early stage of the formation of these DRX grains, some subgrains with an irregular shape in the Region 2 are closed and semi-closed by LAGBs and HAGBs in Fig. 13 (d). These subgrains orientation in the circular area show a relatively wide span in the inverse pole map (Fig. 13(c)), which proves the occurrence of subgrains rotation inside the original deformed grains. In addition, Fig. 13(d) shows that the point to origin profile constantly accumulates misorientations based on measuring the point-to-point misorientation gradient of the A-B line. During the subsequent deformation, through the increasing misorientation caused by progressive rotation of subgrains, these SBs first transition to MAGBs and then gradually change to HAGBs [52], so that DRX grains will be generated inside the initially deformed grains.

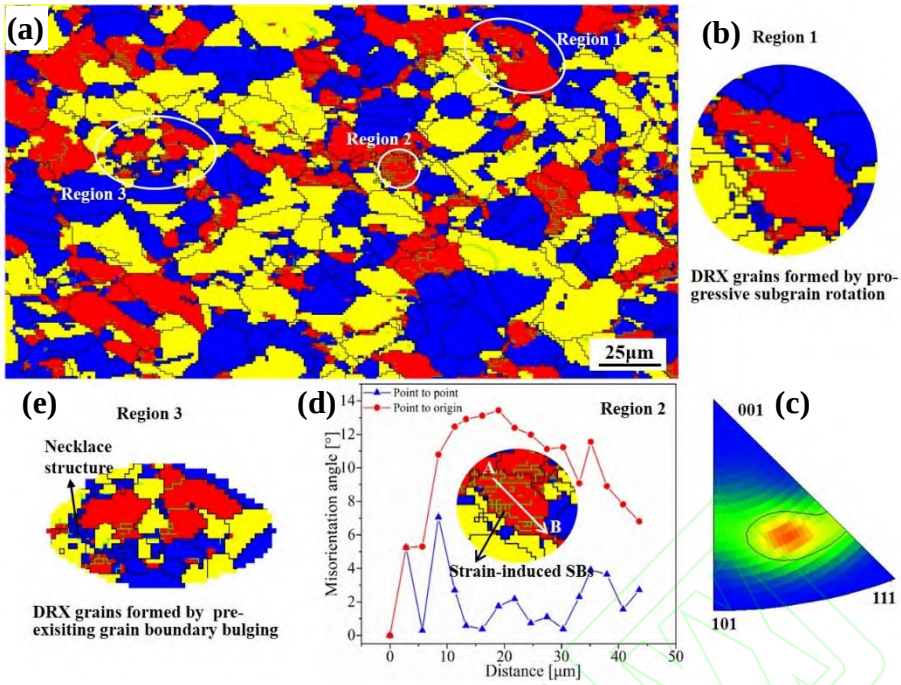

Fig. 13 (a) DRX-distributed map of N6 obtained at 900 °C/1 s^{-1}, (b) the enlarged view of Region 3, (c) the orientation relationships represented by inverse pole map of Region 2, (d) the misorientation gradient of Region 2 measured along the A-B line, (e) the enlarged view of Region 1

MAGBs are regarded to be a transition stage from LAGBs to HAGBs and the role of CDRX can be enhanced through the increased fraction of MAGBs [53,54]. Fig. 14 shows the fractions of different types of GBs after thermal compression at 900 °C and different strain rates. When the strain rate is less than 1 s^{-1}, the proportion of MAGBs increases with increasing strain rate, which indicates that the role of CDRX is enhanced with increasing strain rate. When the strain rate is greater than 1 s^{-1}, the fraction of MAGBs decreases with increasing strain rate, indicating that the role of CDRX weakens. Studies have shown that the effect of CDRX is weakened by the increase in temperature [55]. Therefore, the reason for the weakening of the CDDX effect at 10 s^{-1} is the temperature rise effect in the thermal deformation process with high strain rate. The fraction of MAGBs in Fig. 14 shows that the CDRX mechanism is more easily triggered at 900 °C/1.0 s^{-1}. Although CDRX can be observed under the studied deformation conditions, the smaller fraction of MAGBs observed in the study clearly shows that the incidence of CDRX is the lowest. Meanwhile, the fractions of HAGBs in Fig. 14 all show a considerable dominance at different conditions, which also is

a characteristic of DDRX. Therefore, it is concluded that DDRX is a dominant DRX mechanism in N6, while CDRX is not an active one.

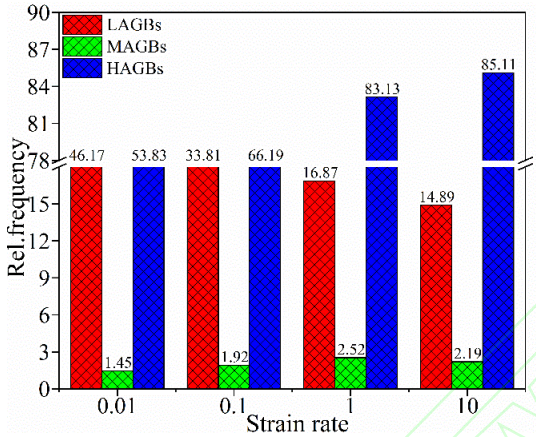

Fig. 14 Fractions of LAGBs, MAGBs and HAGBs at 900 °C and different strain rate

Fig. 15 is the typical TEM micrograph obtained at 900 °C/1 s^{-1}. It can be seen from Fig. 15(a) that a large number of base plane slip bands (yellow lines) exist during the thermal deformation process, which confirms that N6 is a plastic deformation mode dominated by slip, while a large number of dislocation tangles form dislocation cells with different sizes (marked by white dashed line). The dislocation cells will transform into complete subgrains, and subgrains rotate and merge to form a DRX nucleus and then DRX grains form. Such a process is typical of CDRX [6]. During thermal deformation, the distorted twin boundaries can serve as preferential DRX nucleation sites [56]. It was found that TBs occurred via slip deformation, forming a stepped morphology (Fig. 15(b)). Strain fields surrounding the sharp edges of the twin steps give rise to local accumulation of dislocations and the formation of arrays of dislocation walls, which ultimately promote the nucleation of DRX [56]. The TEM images (Fig. 15(c)) confirm the existence of bulging boundaries (marked by the green dashed line). The dislocation accumulation densities located at the bulging boundaries are much greater than that of the straight boundary (marked by the blue dashed line), which further confirms bulging GBs will preferentially become the nucleation site. In addition, some dense dislocation walls (DDWs) (marked by the red dashed line) of a few microns in length were observed (Fig. 15(d)), and the widths of the DDWs are rather narrow. The segmentation of original DDWs is the most common mechanism for the formation of new CBs [57]. A schematic for the above

mentioned DRX mechanisms under the testing conditions for N6 is illustrated in Fig. 16, where (a) shows the CDRX processes, and (b) and (c) show the DDRX processes.

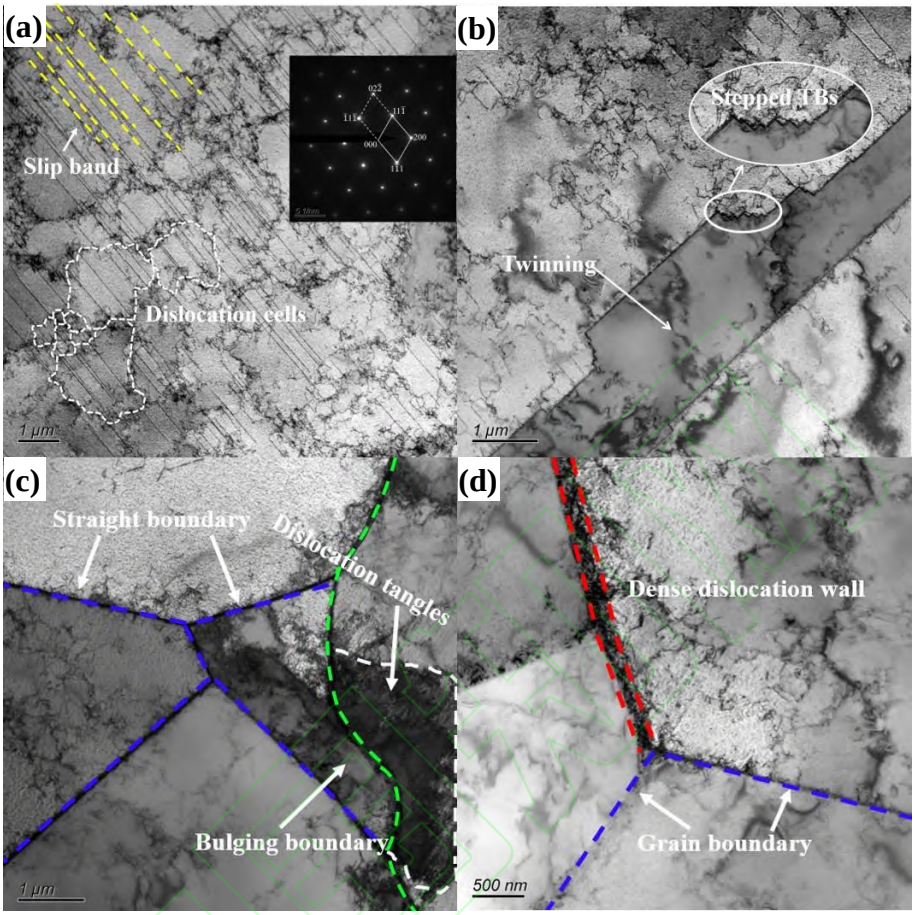

Fig. 15 TEM micrograph of (a) dislocation cells, (b) stepped twin, (c) bulging boundary, (d) DDW in the specimen deformed at 900 °C with a strain rate of 1 s^{-1}

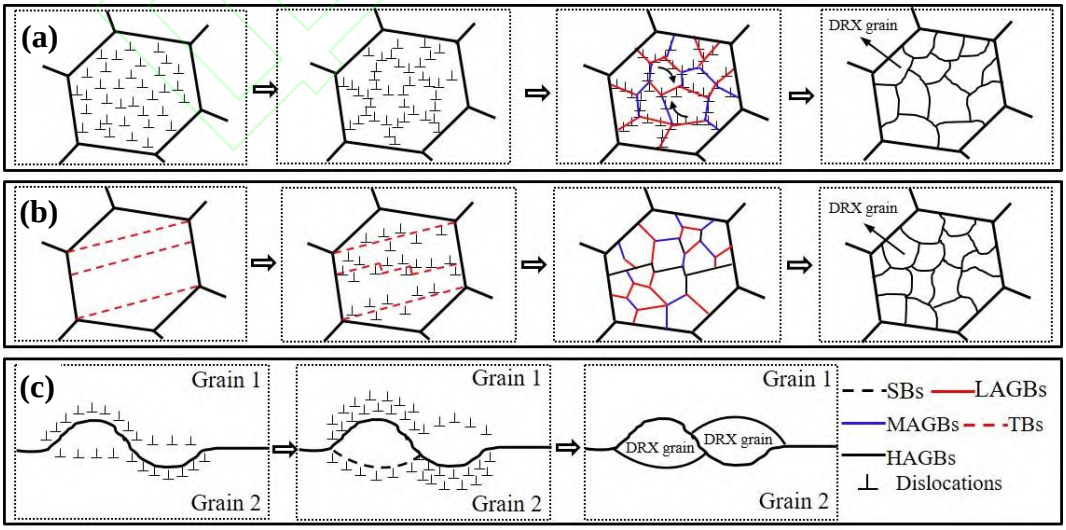

Fig. 16 Schematic of DRX mechanisms: (a) CDRX, (b) and (c) DDRX

5 Conclusions

The hot deformation behavior and DRX mechanism in N6 was studied in the temperature range of 900-1200 ℃ at strain rates of 0.01-10s^{-1} through isothermal compression tests. The primary outcomes of this study are summarized as follows:

(1) The flow behavior of N6 is significantly affected by the strain rate and deformation temperature. An obvious adiabatic temperature rise effect is generated at high strain rate (10 s^{-1}), and the constitutive equation and processing map are built up from the temperature corrected stress-strain curves.

(2) The optimal processing parameters for good workability are obtained in the temperature range of 950-1050 ℃ with a strain rate range of 0.1~1 s^{-1}. Full DRX occurs under optimal conditions.

(3) The DRX process of N6 alloy is a thermally activated process and particularly sensitive to strain rate. At the same temperature, the degree of dynamic recrystallization is gradually strengthened with increasing strain rate. The adiabatic heat effect and higher strain energy and dislocation density serve as vital reasons for the accelerated DRX at high strain rates.

(4) The DRX mechanism of N6 is shown as the dominant nucleation mechanism of DDRX characterized by grain boundary bulging and twin-assisted nucleation, while the CDRX characterized by subgrain combined with rotation is an inactive nucleation mechanism. In addition, the effect of CDRX nucleation strengthens first and then weakens with increasing strain rate.

Acknowledgements

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N6 的热加工性和动态再结晶机制

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摘要: 利用热压缩实验系统地研究了 N6 的热加工性和动态再结晶机制。基于热压缩实验数据，建立了 N6 的本构方程并验证了其可靠性，同时构建了其热加工图并结合显微组织观察建立了热变形参数与微观结构之间的半定量响应关系。 N6 的动态再结晶过程是一个热激活过程，对应变速率特别敏感。N6 的最佳热处理工艺参数确定为 950-1050 ℃/ 0.1~1 s^{-1}。此外，还证明了以晶界弓出和孪晶协助成核为特征的不连续动态再结晶机制为主要的动态再结晶机制，而以亚晶旋转合并为特征的连续动态再结晶是辅助机制。

关键词：微观组织演变，热加工性，加工图，动态再结晶