# Energy levels and transition probabilities of quartet Rydberg series in boronlike silicon 

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#### Abstract

Excited-state energy levels for quartet states $1 s^{2} 2 l 2 l^{\prime} n l^{\prime \prime}\left(\mathrm{n}=2-7 ; l=s, p ; l^{\prime}=s, p ; l^{\prime \prime}=s, p, d, f\right.$ ) pertaining to ${ }^{4} L^{\mathrm{e}, \mathrm{o}}$ (m) ( $L=S, P, D ; \mathrm{m}=1-5$ ) Rydberg series in $\mathrm{Si}^{9+}$ ion are studied, using multi-configuration Rayleigh-Ritz variation method. Corrections for the relativistic effects, mass polarization effect, and quantum electrodynamics (QED) effect are taken into account to obtain the accurate relativistic energies. Transition probabilities, line strengths, oscillator strengths, and wavelengths of electric-dipole transitions between these quartet states are systematically calculated and compared with available reference data. These calculated data could be used for the identification of spectral lines arising from the coronal atmospheres and experimental study in future work.


## 1. Introduction

As the important element in the Universe, silicon is found in the process of stellar nucleosynthesis and supernova explosions [1]. Photoemission lines, in the extreme ultraviolet (EUV) and soft X-ray range, arising from various highly stripped silicon ions play a prominent role in the spectra of astrophysical plasmas from a wide variety of astronomical sources such as the solar corona [2,3], stellar atmospheres [4,5], and supernova remnants [6]. The information from these photoemission lines are often used to diagnose temperature, density [7] and element abundance [8] in astronomical object. During the past decades, considerable progress has been made in this field.

Early in 1970 s, Träbert et al. [9-12] measured the lifetimes and identified the spectral lines from variously ionized silicon in the ranges of vacuum ultraviolet and x-ray, using the beam-foil technique. Later, Faenov [13] observed and measured the wavelengths of satellites due to 41 transitions in boron-like $\mathrm{Si}^{9+}$ ions in $\mathrm{CO}_{2}$ laser-produced plasma. Recently, Bernhardt et al. [14] performed the measurement of electronion recombination coefficients of $\mathrm{Si}^{10+}$ and $\mathrm{Si}^{9+}$ ions, using the elec-tron-ion merged-beams technique at the heavy-ion storage ring.

Theoretically, there are a few calculations of transition probabilities of boron-like $\mathrm{Si}^{9+}$ ion available in the literatures. In the work of the Opacity Project, energy levels, oscillator strengths and photo-ionization
cross sections of multi-excited states in boron-like ions with $\mathrm{Z}=6-16$ were calculated in Ref. [15], by using the configuration interaction code CIV3 and $R$-matrix method. As part work of the IRON Project, radiative rates of the E1, E2 and M1 transitions of $n=2$ complex in the boron isoelectronic sequence ( $Z=8-26$ ) were reported by Galavís et al. [16], using the computer program SUPERSTRUCTURE code. Merkelis et al. [17] calculated the energy spectra and electric dipole transitions in the boron isoelectronic sequence ( $Z=8-26$ ) between the levels of the $\mathrm{n}=2$ complex ( $2 s^{2} 2 p, 2 s 2 p^{2}$ and $2 p^{3}$ ), with the electron correlations are considered by the stationary second-order many-body perturbation theory (MBPT). For the $n=3$ configurations in boron-like ions with $Z=6-30$, Safronova et al. [18] calculated the corresponding energy levels and fine structure separations by relativistic many-body perturbation theory (RMBPT). Then, Vilkas et al. [19] studied the electric dipole transition probabilities for all levels up to $n=4$ configurations of boron-like $\mathrm{Si}^{9+}$ ion, by the relativistic multireference many-body perturbation (MR-MP) theory. Calculations of energy levels and transition data of boron-like $\mathrm{Si}^{9+}$ ion were also performed by the relativistic coupled-cluster theory [20], multi-configuration Hartree-Fock (MCHF) method [21], multi-configuration Dirac-Fock (MCDF) method [22,23], 1/Z perturbation theory method [24], and AUTOSTRUCTURE code [5,25]. At present, the experimental and calculated transition coefficients for these configuration states in $\mathrm{Si}^{9+}$ ion are critically compiled.

[^0]These data can be available from the open NIST atomic spectra database [26] and CHIANTI atomic database [27]. However, most of earlier works mentioned above focused on the low-lying states with $\mathrm{n} \leq 4$. Studies on the $\mathrm{n} \geq 5$ quartet series of $\mathrm{Si}^{9+}$ ion are still fragmentary. Besides, large discrepancies still exist between theoretical data and experimental results, and also between different theoretical methods. Given the increasing need for theoretical spectroscopic data in astrophysics, it is necessary to calculate more accurate atomic data such as the wavelengths and transition probabilities for the Rydberg quartet states of boron-like ions.

With this in mind, in this paper, we have performed systematical calculations of energy levels and radiative transitions of $1 s^{2} 2 l 2 l^{\prime} n l^{\prime \prime} L^{\mathrm{e}, \mathrm{o}}$ (m) $(L=S, P, D ; \mathrm{m}=1-5)$ Rydberg series in the boron-like $\mathrm{Si}^{9+}$ ion. The Rayleigh-Ritz variational results are reported for calculations done by considering the multi-configuration interaction with moderate-scale wave functions and the mass polarization and relativistic effects with the first-order perturbation theory. The QED effect corrections are also included. Further, the line strengths, oscillator strengths, transition probabilities, and transition wavelengths of the electric dipole transitions between centers of gravity of $L S$ terms of these Rydberg series in $\mathrm{Si}^{9+}$ ion are calculated and discussed.

## 2. Theoretical method

For the boron-like $\mathrm{Si}^{9+}$ ion, the non-relativistic Hamiltonian in the $L S$ coupling scheme is
$H_{0}=\sum_{i=1}^{5}\left[-\frac{1}{2} \nabla_{i}^{2}-\frac{Z}{r_{i}}\right]+\sum_{\substack{i, j=1 \\ i<j}}^{5} \frac{1}{r_{i j}}$.
The wave function $\Psi_{b}(1,2,3,4,5)$ is constructed from $L S$ coupled trial functions,
$\Psi_{b}(1,2,3,4,5)=A \sum_{i} C_{i} \phi_{\mathrm{n}(i)}(R) Y_{l(\mathrm{i})}^{\mathrm{LM}}(\Omega) \chi_{S S_{z}}$,
where $A$ is the antisymmetrization operator, and $C_{i}$ is a linear parameter. $R$ represents, collectively, the radial parts of $r_{1}, r_{2}, r_{3}, r_{4}, r_{5}$ and $\Omega$ represents their angular part. $\mathrm{n}(i)$ represents the combinations of the five principal quantum numbers in the $i$ th configuration. $\phi_{\mathrm{n}(\mathrm{i})}(R)$ is the set of radial functions,
$\phi_{n(i)}(R)=\prod_{j=1}^{5} r_{j}^{n_{j}} \exp \left(-\alpha_{j} r_{j}\right)$.
A diff ;erent nonlinear parameter $\alpha_{j}$ is optimized for each electron. $l(i)$ is the orbital angular coupling function in $i$ th configuration. $L$ and $M$ represent the total orbital angular momentum and its azimuthal component, respectively. $Y_{l(\mathrm{i})}^{\mathrm{L} M}(\Omega)$ is the function of angular component and $\chi_{S S_{z}}$ is the function of spin wave function with the total spin $S$.

In this work, all the coupling modes of the orbital and spin angular momenta are considered. The configurations which have the same $Y_{l(\mathrm{i})}^{\mathrm{L} M}(\Omega)$ and $\chi_{S S_{z}}$ are treated as a "partial wave". In each partial wave, the orbital angular coupling function $l(i)$ and spin wave function $\chi_{S S_{z}}$ are fixed, and the principal quantum numbers $\mathrm{n}_{j}(j=1-5)$ of electrons increase to expand configuration states functions (CSFs). If the maximum value of each $n_{j}$ is given, the number of CSFs in each partial wave is restricted by
$\sum_{j} \mathrm{n}_{j} \leq b$.
Here $b$ is an adjustable parameter. The set of CSFs in each partial wave expands until the convergence of the expectation value of the energies is obtained. Since all the principal quantum numbers of electrons are treated as "active", the valence-valence, core-valence, and core-core correlations could be considered in this work. For the low-lying quartet Rydberg series configuration states, the significant configuration
interactions arise from the $\mathrm{n} \leq 5$ layers. While for the high-lying quartet Rydberg series configuration states, the configuration space expands to $\mathrm{n}=9$.

The variation of the non-relativistic energy $E_{b}$ with respect to the linear parameters $C_{i}$ leads to a secular equation whose eigenvalue is a function of the nonlinear parameters $\alpha_{j}$. Then the $E_{b}$ and wave functions are obtained via the Rayleigh-Ritz variation calculation by
$\delta\left[\frac{\left\langle\Psi_{b}\right| H_{0}\left|\Psi_{b}\right\rangle}{\left\langle\Psi_{b} \mid \Psi_{b}\right\rangle}\right]=0$.
To saturate the wave functional space and further improve the nonrelativistic energy $E_{b}$, the restricted variational method is used to calculate the restricted variational energy contributions $\Delta E_{R V}$. Then the non-relativistic energies are obtained by $E_{b}+\Delta E_{R V}$. The detailed explanations of the restricted variational calculation were given in Refs. [28,29], and we will not repeat here.

Accurate transition wavelengths demand relativistic energy eigenvalues. Here, the correction $\Delta E_{m p}+\Delta E_{\text {rel }}$ for the mass polarization effect and the relativistic effects are also considered by using the firstorder perturbation theory. The relativistic effects considered in this work are the correction to the kinetic energy, the Darwin term, the electron-electron contact term, and the orbit-orbit interaction term. To estimate QED effect for these states, the screened hydrogenic formula [30] is used. The detailed expressions for the correction $\Delta E_{Q E D}$ have been discussed in Ref. [31]. The total correction $\Delta E_{\text {corr }}$ is obtained by $\Delta E_{\text {corr }}=\Delta E_{m p}+\Delta E_{\text {rel }}+\Delta E_{Q E D}$. Then the centers of gravity of energy is obtained by
$E_{\text {total }}=E_{b}+\Delta E_{R V}+\Delta E_{\text {corr }}$.
For an electric dipole transition, the transition line strength $S$ (in a.u.) is calculated by
$S=\left|<\psi_{f}\|\boldsymbol{D}\| \psi_{i}>\right|^{2}$,
where $\boldsymbol{D}$ is the electric dipole transition operator, $\psi_{f}$ and $\psi_{i}$ are the nonrelativistic wave functions of the final and initial configuration states, respectively. The relationship [32] between electric dipole transition probability $A_{i k}$, absorption oscillator strength $f_{k i}$, and line strength $S$ is,
$A_{i k}=\frac{6.6702 \times 10^{15}}{\lambda^{2}} \frac{g_{k}}{g_{i}} f_{k i}=\frac{2.0261 \times 10^{18}}{\lambda^{3} g_{i}} S$.
Here, $\lambda$ (in $\AA$ ) is the transition wavelength between the initial and final configuration states. The statistical weights $g_{i}$ and $g_{k}$ are obtained by
$g_{i(k)}=\left(2 L_{i(k)}+1\right)\left(2 S_{i(k)}+1\right)$
In this work, two expressions (the length form and velocity form) for the oscillator strength and transition probability are given. Ideally, these two forms of the same quantity should agree, so the discrepancies remaining between them can reflect the accuracy of theoretical calculation.

## 3. Results and discussion

In this work, moderate-scale multi-configuration wave functions are used to calculate the energies and wave functions of the quartet ${ }^{4} L^{\mathrm{e}, \mathrm{o}}$ (m) $(L=S, P, D ; \mathrm{m}=1-5)$ Rydberg series in boron-like $\mathrm{Si}^{9+}$ ion. In the energy calculation of these quartet states, the radial and angular correlations are very important, especially for the ${ }^{4} D^{\mathrm{e}, \mathrm{o}}$ states. For the even-parity complexes, the important orbital angular series $\left[l_{1}, l_{2}, l_{3}, l_{4}, l_{5}\right] \quad$ are $[0,0,0, l, l], \quad[0,0,0, l, l+2], \quad[0,0,1, l, l+1]$, $[0,0,1, l, l+3], \quad[0,0,2, l, l], \quad[0,1,1, l, l], \quad[0,1,1, l, l+2]$, $[1,1,1, l, l+1]$, etc. For the odd-parity complexes, the important orbital angular series are $[0,0,0, l, l+1],[0,0,1, l, l],[0,0,1, l, l+2]$, $[0,0,2, l, l+1], \quad[0,1,1, l, l+1], \quad[0,1,2, l, l], \quad[1,1,1, l, l]$, $[1,1,1, l, l+2]$, etc. In both cases, the value of $l$ is from 0 to 8 , as the

Table 1
Centers of gravity of $L S$ terms (a.u.) of quartet ${ }^{4} L^{\mathrm{e}, \mathrm{o}}(\mathrm{m})(L=S, P, D ; \mathrm{m}=1-5)$ Rydberg series in $\mathrm{Si}^{9+}$ ion. $E_{b}+\Delta E_{R V}$ is the nonrelativistic energy, $\Delta E_{\text {corr }}$ are corrections to the mass polarization, relativistic, and QED effects, and $E_{\text {total }}$ is the total energy. $E_{S}$ (Ry) is the energy level relative to the $2 s^{2} 2 p{ }^{2} p^{\circ}$ ground term.

| Configuration states | $E_{b}+\Delta E_{R V}$ | $\Delta E_{\text {corr }}$ | $E_{\text {total }}$ | $E_{S}(\mathrm{Ry})$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | Present | NIST | MR-MP | AS |
| $2 s 2 p^{2}{ }^{4} P^{\mathrm{e}}(1)$ | -238.06445 | -0.49280 | -238.55725 | 1.4590 | 1.4599 | 1.4582 | 1.4599 |
| $2 s 2 p 3 p{ }^{4} P^{\mathrm{e}}$ (2) | -229.26089 | -0.48695 | -229.74784 | 19.0779 |  | 19.0761 | 19.0655 |
| $2 p^{2} 3 s^{4} P^{\mathrm{e}}(3)$ | -228.41526 | $-0.45910$ | -228.87436 | 20.8248 |  | 20.8150 | 20.8300 |
| $2 p^{2} 3 d^{4} P^{\mathrm{e}}$ (4) | -227.71286 | -0.44819 | -228.16105 | 22.2514 | 22.24 | 22.2437 | 22.3123 |
| $2 s 2 p 4 p{ }^{4} P^{\mathrm{e}}(5)$ | -226.57325 | -0.48193 | -227.05518 | 24.4632 |  | 24.3623 | 24.5152 |
| $2 s 2 p 3 s^{4} P^{\circ}(1)$ | -229.71401 | $-0.49528$ | -230.20929 | 18.1550 | 18.1553 | 18.1537 | 18.1470 |
| $2 s 2 p 3 d^{4} P^{\circ}(2)$ | -228.97960 | -0.48254 | -229.46214 | 19.6493 | 19.654 | 19.6543 | 19.6778 |
| $2 p^{2} 3 p^{4} P^{\circ}(3)$ | -228.07832 | -0.45381 | -228.53213 | 21.5093 |  | 21.4912 | 21.5320 |
| $2 s 2 p 4 s^{4} P^{\circ}(4)$ | -226.73852 | -0.48391 | -227.22243 | 24.1287 |  | 24.1202 | 24.1610 |
| $2 s 2 p 4 d^{4} P^{\circ}(5)$ | -226.46405 | -0.47932 | -226.94337 | 24.6868 |  | 24.6896 | 24.7491 |
| $2 s 2 p 3 p^{4} D^{\mathrm{e}}(1)$ | -229.38261 | -0.48824 | -229.87085 | 18.8318 |  | 18.8371 | 18.8358 |
| $2 p^{2} 3 d^{4} D^{\mathrm{e}}(2)$ | -227.76768 | -0.44664 | -228.21432 | 22.1449 |  | 22.0664 |  |
| $2 s 2 p 4 p^{4} D^{\mathrm{e}}(3)$ | -226.61094 | $-0.48586$ | -227.09680 | 24.3799 |  | 24.4393 | 24.4329 |
| $2 s 2 p 4 f^{4} D^{\mathrm{e}}$ (4) | -226.39004 | $-0.48240$ | -226.87244 | 24.8287 |  | 24.8153 | 24.8865 |
| $2 s 2 p 5 p^{4} D^{\mathrm{e}}(5)$ | -225.37523 | -0.47962 | -225.85485 | 26.8638 |  |  |  |
| $2 s 2 p 3 d^{4} D^{\circ}(1)$ | -229.01712 | -0.48423 | -229.50135 | 19.5708 | 19.570 | 19.5708 | 19.6069 |
| $2 p^{2} 3 p^{4} D^{\circ}(2)$ | -228.11632 | -0.45398 | -228.57030 | 21.4329 |  | 21.4116 | 21.4468 |
| $2 s 2 p 4 d^{4} D^{\circ}(3)$ | -226.47517 | $-0.48311$ | -226.95828 | 24.6570 |  | 24.6582 | 24.7103 |
| $2 p^{2} 4 p^{4} D^{\circ}(4)$ | -225.37636 | $-0.45064$ | -225.82700 | 26.9195 |  | 26.9114 |  |
| $2 s 2 p 5 d^{4} D^{\circ}(5)$ | -225.31052 | -0.48062 | -225.79114 | 26.9913 |  |  |  |
| $2 s 2 p 3 p{ }^{4} S^{\mathrm{e}}(1)$ | -229.31907 | -0.48525 | -229.80432 | 18.9649 |  | 18.9613 | 18.9609 |
| $2 s 2 p 4 p{ }^{4} \mathrm{~S}^{\mathrm{e}}(2)$ | -226.59175 | -0.48503 | -227.07678 | 24.4200 |  | 24.4233 | 24.4791 |
| $2 s 2 p 5 p{ }^{4} S^{\mathrm{e}}(3)$ | -225.37095 | -0.48280 | -225.85375 | 26.8660 |  |  |  |
| $2 s 2 p 6 p{ }^{4} S^{\text {e }}$ (4) | -224.71671 | $-0.48010$ | -225.19681 | 28.1799 |  |  |  |
| $2 s 2 p 7 p{ }^{4} S^{\mathrm{e}}(5)$ | -224.30469 | -0.48083 | -224.78552 | 29.0025 |  |  |  |
| $2 p^{34} S^{\circ}(1)$ | -236.52523 | $-0.45974$ | -236.98497 | 4.6036 | 4.5989 | 4.5976 | 4.5985 |
| $2 p^{2} 3 p^{4} S^{\circ}(2)$ | -227.95663 | -0.45318 | -228.40981 | 21.7539 |  | $21.7393$ | 21.8557 |
| $2 p^{2} 4 p^{4} S^{\circ}(3)$ | -225.33190 | $-0.45063$ | -225.78253 | 27.0085 |  | 27.0088 |  |
| $2 p^{2} 5 p^{4} S^{\circ}(4)$ | -224.12740 | -0.44964 | -224.57704 | 29.4195 |  |  |  |
| $2 p^{2} 6 p^{4} S^{\circ}(5)$ | -223.47932 | -0.44831 | $-223.92763$ | 30.7183 |  |  |  |

NIST: Ref. [26]. MR-MP: multireference many-body perturbation, Ref. [19]. AS: AUTOSTRUCTURE, Refs. [25,27].


Fig. 1. Deviations of the calculated term center-of-gravity values from the reference results of the ${ }^{4} L^{\mathrm{e}, \mathrm{o}}(L=S, P, D)$ Rydberg series in boron-like $\mathrm{Si}^{9+}$ ion.
energy contribution from the set with $l>8$ is less than $10^{-5}$ a.u. and can be neglected. For an orbital angular momentum set $\left[l_{1}, l_{2}, l_{3}, l_{4}, l_{5}\right]$, all kinds of coupling modes which are possible to obtain the desired total orbital angular momentum $L$ are considered. For the quartet state, four spin angular coupling modes are possible, namely,
$\chi^{1}=\left\{\left[\left(\frac{1}{2}, \frac{1}{2}\right) 0, \frac{1}{2}\right] \frac{1}{2}, \frac{1}{2}\right\} 1, \frac{1}{2} ; \chi^{2}=\left\{\left[\left(\frac{1}{2}, \frac{1}{2}\right) 1, \frac{1}{2}\right] \frac{1}{2}, \frac{1}{2}\right\} 1, \frac{1}{2}$
$\chi^{3}=\left\{\left[\left(\frac{1}{2}, \frac{1}{2}\right) 1, \frac{1}{2}\right] \frac{3}{2}, \frac{1}{2}\right\} 1, \frac{1}{2} ; \chi^{4}=\left\{\left[\left(\frac{1}{2}, \frac{1}{2}\right) 1, \frac{1}{2}\right] \frac{3}{2}, \frac{1}{2}\right\} 2, \frac{1}{2}$
The partial waves consist of all combinations of orbital-spin angular momentum coupling modes. For accurate calculation, all coupling
modes are considered here, except for the partial waves with the energy contributions less than $10^{-5}$ a.u. To reach a high accuracy, more than 110 partial waves and 4000 configuration terms are used to construct high-quality wave functions in this work.

In Table 1, we list the calculated non-relativistic energies $E_{b}+\Delta E_{R V}$, corrections $\Delta E_{\text {corr }}$ to the mass polarization effect, relativistic effects and QED effect, and total energy levels $E_{\text {total }}$ of the quartet Rydberg series ${ }^{4} L^{\mathrm{e}, \mathrm{o}}(\mathrm{m})(L=S, P, D ; \mathrm{m}=1-5)$ in $\mathrm{Si}^{9+}$ ion. The configurations of these Rydberg series are listed and numbered according to the values of energy levels. For the convenience of comparison, the relative energy levels $E_{S}$ (with respect to the ground term $2 s^{2} 2 p^{2} P^{0}$ ) are also given in Table 1, together with 6 NIST experimental values [26], the AS calculations [25] from the CHIANTI database [27], and theoretical data from MR-MP theory calculations [19]. In our calculations, the contribution from the finite nuclear size correction is not considered. According to the analysis and calculation in Refs. [33-35], the finite nuclear size effect to the total energy levels is estimated to be about 0.00005 a.u... Combining the finite nuclear size effect and the omitted high-l configuration interactions, the estimation of uncertainty of calculated energy level is about 0.00015 a.u. Fig. 1 gives the deviations of the calculated term center-of-gravity values from the reference results for these Rydberg series in $\mathrm{Si}^{9+}$ ion. As can be seen from Fig. 1, our calculated values show good agreement with the values from NIST, due to large amount of electron correlation effects considered in this work. The root mean square (RMS) deviation between our results and NIST values is about 0.0054 Ry. Overall, our results show better agreement with the MR-MP theoretical data [19] than those from AS calculations [25], due to limited correlations between valence electrons are considered in the AS calculations. The RMS deviations between our calculations and those from MR-MP and AS are 0.0299 Ry and 0.0412 Ry, respectively. Large differences still exist between the theoretical data for these Rydberg

Table 2
Line strengths in length form $S_{P}$ (a.u.), oscillator strengths $f_{k i}$ (in the length $f_{l}$ and velocity $f_{v}$ ), transition probabilities $A_{i k}$ (in the length $A_{l}$ and velocity $A_{v}$ ) (s ${ }^{-1}$ ), and vacuum wavelengths $\lambda(\AA)$ for electric dipole transitions between the ${ }^{4} L^{\text {e,o }}(L=S, P, D)$ Rydberg series terms in $\mathrm{Si}^{9+}$ ion. Only the transitions with $S_{P}>3 \times 10^{-5}$ (a.u.) are listed. aEb means $\mathrm{a} \times 10^{\mathrm{b}}$.

| Upper level | Lower level | $S_{P}$ |  | $A_{i k}$ |  |  | $f_{k i}$ |  |  | $\lambda(\AA)$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Present | Others | $A_{l}$ | $A_{v}$ | Others | $f_{l}$ | $f_{v}$ | Others | Present | Others |
| $2 p^{2} 6 p^{4} S^{\circ}$ | $2 s 2 p^{2}{ }^{4} p^{\text {e }}$ | $2.27 \mathrm{E}-03$ |  | $3.80 \mathrm{E}+10$ | $3.79 \mathrm{E}+10$ |  | $1.84 \mathrm{E}-03$ | $1.84 \mathrm{E}-03$ |  | 31.15 |  |
| $2 p^{2} 5 p^{4} S^{\circ}$ | $2 s 2 p^{2}{ }^{4} p^{\text {e }}$ | 4.39E-03 |  | $6.42 \mathrm{E}+10$ | $6.40 \mathrm{E}+10$ |  | $3.41 \mathrm{E}-03$ | $3.41 \mathrm{E}-03$ |  | 32.59 |  |
| $2 p^{2} 4 p^{4} S^{\circ}$ | $2 s 2 p^{24} p^{\text {e }}$ | $1.09 \mathrm{E}-02$ |  | $1.21 \mathrm{E}+11$ | $1.21 \mathrm{E}+11$ |  | $7.72 \mathrm{E}-03$ | $7.67 \mathrm{E}-03$ |  | 35.67 |  |
| $2 s 2 p 5 d^{4} D^{\circ}$ | $2 s 2 p^{2}{ }^{4} p^{\text {e }}$ | 6.60E-02 |  | $1.47 \mathrm{E}+11$ | $1.46 \mathrm{E}+11$ |  | $4.68 \mathrm{E}-02$ | $4.64 \mathrm{E}-02$ |  | 35.69 |  |
| $2 p^{2} 4 p{ }^{4} D^{\circ}$ | $2 s 2 p^{24} p^{\text {e }}$ | $4.45 \mathrm{E}-02$ |  | $9.83 \mathrm{E}+10$ | $9.73 \mathrm{E}+10$ |  | $3.15 \mathrm{E}-02$ | $3.12 \mathrm{E}-02$ |  | 35.79 |  |
| $2 \mathrm{~s} 2 p 4{ }^{4} \mathrm{P}^{\circ}$ | $2 s 2 p^{2}{ }^{4} p^{\text {e }}$ | $1.04 \mathrm{E}-01$ |  | $2.90 \mathrm{E}+11$ | $2.88 \mathrm{E}+11$ | $2.52 \mathrm{E}+11^{\text {a }}$ | $6.70 \mathrm{E}-02$ | $6.64 \mathrm{E}-02$ | $5.79 \mathrm{E}-02^{\text {a }}$ | 39.23 | $39.13^{\text {a }}$ |
| $2 s 2 p 4 d^{4} D^{\circ}$ | $2 s 2 p^{2}{ }^{4} p^{\text {e }}$ | $3.02 \mathrm{E}-01$ |  | $5.05 \mathrm{E}+11$ | $5.01 \mathrm{E}+11$ | $4.21 \mathrm{E}+11^{\text {a }}$ | $1.95 \mathrm{E}-01$ | $1.93 \mathrm{E}-01$ | $1.61 \mathrm{E}-01^{\text {a }}$ | 39.28 |  |
| $2 s 2 p 4 s^{4} p^{\text {o }}$ | $2 s 2 p^{2}{ }^{4} p^{\text {e }}$ | $1.99 \mathrm{E}-02$ |  | $5.18 \mathrm{E}+10$ | $5.18 \mathrm{E}+10$ | $3.82 \mathrm{E}+10^{\text {a }}$ | $1.25 \mathrm{E}-02$ | $1.26 \mathrm{E}-02$ | $9.22 \mathrm{E}-03^{\text {a }}$ | 40.20 | $40.14^{\text {a }}$ |
| $2 p^{2} 3 p^{4} S^{\circ}$ | $2 s 2 p^{24} p^{\text {e }}$ | $4.41 \mathrm{E}-02$ |  | $2.47 \mathrm{E}+11$ | $2.44 \mathrm{E}+11$ | $2.33 \mathrm{E}+11^{\text {a }}$ | $2.49 \mathrm{E}-02$ | $2.46 \mathrm{E}-02$ | $2.33 \mathrm{E}-02^{\text {a }}$ | 44.90 | $44.93{ }^{\text {a }}$ |
| $2 p^{2} 3 p{ }^{4} p^{\text {o }}$ | $2 s 2 p^{2}{ }^{4} p^{\text {e }}$ | $9.85 \mathrm{E}-02$ |  | $1.77 \mathrm{E}+11$ | $1.79 \mathrm{E}+11$ | $1.77 \mathrm{E}+11^{\text {a }}$ | $5.49 \mathrm{E}-02$ | $5.53 \mathrm{E}-02$ | $5.48 \mathrm{E}-02^{\text {a }}$ | 45.45 | $45.40^{\text {a }}$ |
| $2 p^{2} 3 p{ }^{4} D^{\text {o }}$ | $2 s 2 p^{24} p^{\text {e }}$ | $1.24 \mathrm{E}-01$ |  | $1.33 \mathrm{E}+11$ | $1.33 \mathrm{E}+11$ | $1.28 \mathrm{E}+11^{\text {a }}$ | $6.89 \mathrm{E}-02$ | $6.91 \mathrm{E}-02$ | $6.27 \mathrm{E}-02^{\text {a }}$ | 45.62 |  |
| $2 s 2 p 4 p{ }^{4} P^{\text {e }}$ | $2 p^{3}{ }^{\text {S }}$ | 6.64E-04 |  | $1.16 \mathrm{E}+09$ | $1.12 \mathrm{E}+09$ | $1.01 \mathrm{E}+09^{\text {a }}$ | $1.10 \mathrm{E}-03$ | $1.06 \mathrm{E}-03$ | $9.55 \mathrm{E}-04^{\text {a }}$ | 45.89 | $45.76{ }^{\text {a }}$ |
| $2 \mathrm{~s} 2 \mathrm{p} 3 d^{4} \mathrm{P}^{\text {o }}$ | $2 s 2 p^{24} p^{\text {e }}$ | $5.99 \mathrm{E}-01$ |  | $8.04 \mathrm{E}+11$ | $8.02 \mathrm{E}+11$ | $7.89 \mathrm{E}+11^{\text {a }}$ | $3.02 \mathrm{E}-01$ | $3.02 \mathrm{E}-01$ | $2.96 \mathrm{E}-01^{\text {a }}$ | 50.10 | $50.07^{\text {a }}$ |
| $2 s 2 p 3 d^{4} D^{\circ}$ | $2 s 2 p^{2}{ }^{4} \mathrm{p}^{\mathrm{e}}$ | $1.89 \mathrm{E}+00$ | $1.80 \mathrm{E}+00^{\text {b }}$ | $1.50 \mathrm{E}+12$ | $1.50 \mathrm{E}+12$ | $1.43 \mathrm{E}+12^{\text {a }}$ | $9.51 \mathrm{E}-01$ | $9.49 \mathrm{E}-01$ | $8.99 \mathrm{E}-01^{\text {a }}$ | 50.31 | $50.319^{\text {b }}$ |
|  |  |  |  |  |  | $1.43 \mathrm{E}+12^{\text {b }}$ |  |  | $9.05 \mathrm{E}-01^{\text {b }}$ |  |  |
| $2 p^{2} 3 d^{4} p^{\text {e }}$ | $2 p^{3}{ }^{\text {S }}$ - | $1.14 \mathrm{E}+00$ |  | $1.40 \mathrm{E}+12$ | $1.40 \mathrm{E}+12$ |  | $1.68 \mathrm{E}+00$ | $1.68 \mathrm{E}+00$ |  | 51.64 |  |
| $2 s 2 p 3 s{ }^{4} P^{\circ}$ | $2 s 2 p^{2}{ }^{4} p^{\text {e }}$ | $1.87 \mathrm{E}-01$ | $1.89 \mathrm{E}-01^{\text {b }}$ | $1.94 \mathrm{E}+11$ | $1.94 \mathrm{E}+11$ | $1.87 \mathrm{E}+11^{\text {a }}$ | 8.65E-02 | $8.67 \mathrm{E}-02$ | $8.36 \mathrm{E}-02^{\text {a }}$ | 54.58 | $54.58{ }^{\text {a }}$ |
|  |  |  | $1.89 \mathrm{E}-01^{\text {c }}$ |  |  | $1.96 \mathrm{E}+11^{\text {b }}$ |  |  | $8.75 \mathrm{E}-02^{\text {b }}$ |  | $54.582^{\text {b }}$ |
|  |  |  |  |  |  |  |  |  | $8.75 \mathrm{E}-0{ }^{\text {c }}$ |  | $54.56{ }^{\text {c }}$ |
| $2 p^{2} 3 s^{4} p^{\text {e }}$ | $2 p^{3}{ }^{\text {S }}{ }^{\circ}$ | 1.08E-01 |  | $1.02 \mathrm{E}+11$ | $1.02 \mathrm{E}+11$ | $1.00 \mathrm{E}+11^{\text {a }}$ | $1.45 \mathrm{E}-01$ | $1.45 \mathrm{E}-01$ | $1.42 \mathrm{E}-01^{\text {a }}$ | 56.18 | $56.14{ }^{\text {a }}$ |
| $2 s 2 p 3 p{ }^{4} p^{\text {e }}$ | $2 p^{3}{ }^{\text {c }}$ | $9.89 \mathrm{E}-04$ |  | $6.69 \mathrm{E}+08$ | $6.50 \mathrm{E}+08$ | $7.77 \mathrm{E}+08^{\text {a }}$ | $1.19 \mathrm{E}-03$ | $1.16 \mathrm{E}-03$ | $1.38 \mathrm{E}-03^{\text {a }}$ | 62.96 | $62.99^{\text {a }}$ |
| $2 s 2 p 7 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 3 s{ }^{4} \mathrm{p}^{\circ}$ | 7.81E-03 |  | $6.68 \mathrm{E}+09$ | $6.22 \mathrm{E}+09$ |  | $2.35 \mathrm{E}-03$ | $2.19 \mathrm{E}-03$ |  | 84.01 |  |
| $2 s 2 p 6 p{ }^{4} S^{\text {e }}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 s^{4} \mathrm{P}^{\text {o}}$ | $1.41 \mathrm{E}-02$ |  | $9.51 \mathrm{E}+09$ | $9.85 \mathrm{E}+09$ |  | 3.93E-03 | $4.07 \mathrm{E}-03$ |  | 90.90 |  |
| $2 p^{2} 6 p^{4} S^{\circ}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | $1.51 \mathrm{E}-02$ |  | $9.80 \mathrm{E}+09$ | $9.83 \mathrm{E}+09$ |  | $4.15 \mathrm{E}-03$ | $4.17 \mathrm{E}-03$ |  | 92.11 |  |
| $2 s 2 p 7 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 3 d^{4} P^{\circ}$ | $2.65 \mathrm{E}-03$ |  | $1.45 \mathrm{E}+09$ | $1.59 \mathrm{E}+09$ |  | 6.89E-04 | $7.54 \mathrm{E}-04$ |  | 97.43 |  |
| $2 s 2 p 5 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 3 s{ }^{4}{ }^{\circ}$ | 3.59E-02 |  | $1.59 \mathrm{E}+10$ | $1.51 \mathrm{E}+10$ |  | $8.68 \mathrm{E}-03$ | $8.25 \mathrm{E}-03$ |  | 104.61 |  |
| $2 s 2 p 5 p^{4} D^{\text {e }}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 s^{4} \mathrm{p}^{\circ}$ | $2.04 \mathrm{E}-01$ |  | $1.81 \mathrm{E}+10$ | $1.74 \mathrm{E}+10$ |  | $4.94 \mathrm{E}-02$ | $4.77 \mathrm{E}-02$ |  | 104.64 |  |
| $2 p^{2} 5 p{ }^{4} S^{\circ}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | 3.72E-02 |  | $1.58 \mathrm{E}+10$ | $1.53 \mathrm{E}+10$ |  | 8.88E-03 | $8.61 \mathrm{E}-03$ |  | 106.03 |  |
| $2 s 2 p 6 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 3 d^{4} P^{\circ}$ | 5.33E-03 |  | $2.21 \mathrm{E}+09$ | $2.22 \mathrm{E}+09$ |  | $1.26 \mathrm{E}-03$ | $1.26 \mathrm{E}-03$ |  | 106.82 |  |
| $2 p^{2} 6 p{ }^{4} S^{\circ}$ | $2 p^{2} 3 d^{4} p^{\text {e }}$ | 3.10E-03 |  | $1.26 \mathrm{E}+09$ | $1.19 \mathrm{E}+09$ |  | 7.29E-04 | $6.88 \mathrm{E}-04$ |  | 107.63 |  |
| $2 s 2 p 5 d^{4} D^{\circ}$ | $2 s 2 p 3 p{ }^{4} D^{\text {e }}$ | $1.24 \mathrm{E}-01$ |  | $9.04 \mathrm{E}+09$ | $9.06 \mathrm{E}+09$ |  | $1.69 \mathrm{E}-02$ | $1.69 \mathrm{E}-02$ |  | 111.69 |  |
| $2 s 2 p 5 d^{4} D^{\circ}$ | $2 s 2 p 3 p^{4} p^{\text {e }}$ | $3.96 \mathrm{E}-01$ |  | $2.63 \mathrm{E}+10$ | $2.55 \mathrm{E}+10$ |  | $8.71 \mathrm{E}-02$ | $8.45 \mathrm{E}-02$ |  | 115.16 |  |
| $2 s 2 p 7 p^{4} S^{\text {e }}$ | $2 p^{2} 3 p^{4} p^{\text {o }}$ | $3.88 \mathrm{E}-05$ |  | $1.09 \mathrm{E}+07$ | $1.17 \mathrm{E}+07$ |  | $8.07 \mathrm{E}-06$ | $8.65 \mathrm{E}-06$ |  | 121.61 |  |
| $2 s 2 p 5 p^{4} D^{\text {e }}$ | $2 s 2 p 3 d^{4} D^{\circ}$ | $4.56 \mathrm{E}-03$ |  | $2.37 \mathrm{E}+08$ | $2.58 \mathrm{E}+08$ |  | 5.54E-04 | $6.03 \mathrm{E}-04$ |  | 124.95 |  |
| $2 s 2 p 5 p^{4} S^{\text {e }}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 \mathrm{~d}^{4} \mathrm{P}^{\text {o}}$ | $1.67 \mathrm{E}-02$ |  | $4.19 \mathrm{E}+09$ | $4.48 \mathrm{E}+09$ |  | $3.34 \mathrm{E}-03$ | $3.57 \mathrm{E}-03$ |  | 126.27 |  |
| $2 s 2 p 5 p^{4} D^{\text {e }}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 d^{4} \mathrm{P}^{\text {o }}$ | $2.43 \mathrm{E}-02$ |  | $1.22 \mathrm{E}+09$ | $1.16 \mathrm{E}+09$ |  | $4.87 \mathrm{E}-03$ | $4.62 \mathrm{E}-03$ |  | 126.31 |  |
| $2 p^{2} 5 p^{4} S^{\circ}$ | $2 p^{2} 3 d^{4} p^{\text {e }}$ | $9.27 \mathrm{E}-03$ |  | $2.28 \mathrm{E}+09$ | $2.21 \mathrm{E}+09$ |  | $1.85 \mathrm{E}-03$ | $1.78 \mathrm{E}-03$ |  | 127.13 |  |
| $2 s 2 p 4 f^{4} D^{\text {e }}$ | $2 s 2 p 3 s{ }^{4} p^{\circ}$ | $2.34 \mathrm{E}-02$ |  | $9.31 \mathrm{E}+08$ | $9.62 \mathrm{E}+08$ | $5.57 \mathrm{E}+08^{\text {a }}$ | $4.34 \mathrm{E}-03$ | $4.48 \mathrm{E}-03$ | $2.54 \mathrm{E}-03^{\text {a }}$ | 136.55 |  |
| $2 s 2 p 6 p{ }^{4} S^{\text {e }}$ | $2 p^{2} 3 p^{4} p^{\circ}$ | 5.09E-05 |  | $1.01 \mathrm{E}+07$ | $1.90 \mathrm{E}+07$ |  | $9.42 \mathrm{E}-06$ | $1.78 \mathrm{E}-05$ |  | 136.61 |  |
| $2 s 2 p 4 p{ }^{4} \mathrm{P}^{\text {e }}$ | $2 s 2 p 3 s^{4} p^{\circ}$ | $4.51 \mathrm{E}-01$ |  | $2.52 \mathrm{E}+10$ | $2.64 \mathrm{E}+10$ | $2.57 \mathrm{E}+10^{\mathrm{a}}$ | $7.90 \mathrm{E}-02$ | $8.25 \mathrm{E}-02$ | $7.93 \mathrm{E}-02^{\text {a }}$ | 144.46 | $143.10^{\text {a }}$ |
| $2 s 2 p 4 p{ }^{4} S^{e}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 \mathrm{~s}^{4} \mathrm{p}^{\circ}$ | $1.53 \mathrm{E}-01$ |  | $2.52 \mathrm{E}+10$ | $2.54 \mathrm{E}+10$ | $1.52 \mathrm{E}+10^{\text {a }}$ | $2.66 \mathrm{E}-02$ | $2.69 \mathrm{E}-02$ | $1.56 \mathrm{E}-02^{\text {a }}$ | 145.46 | $143.91{ }^{\text {a }}$ |
| $2 s 2 p 4 p{ }^{4} D^{\text {e }}$ | $2 s 2 p 3 s{ }^{4} p^{0}$ | 8.88E-01 |  | $2.87 \mathrm{E}+10$ | $2.88 \mathrm{E}+10$ | $2.96 \mathrm{E}+10^{\text {a }}$ | $1.54 \mathrm{E}-01$ | $1.54 \mathrm{E}-01$ | $1.56 \mathrm{E}-01^{\text {a }}$ | 146.39 |  |
| $2 p^{2} 4 p^{4} S^{\circ}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | $1.56 \mathrm{E}-01$ |  | $2.47 \mathrm{E}+10$ | $2.52 \mathrm{E}+10$ |  | $2.68 \mathrm{E}-02$ | $2.73 \mathrm{E}-02$ |  | 147.37 |  |
| $2 s 2 p 5 d^{4} D^{\circ}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | 9.97E-02 |  | $3.13 \mathrm{E}+09$ | $3.13 \mathrm{E}+09$ |  | $1.71 \mathrm{E}-02$ | $1.71 \mathrm{E}-02$ |  | 147.78 |  |
| $2 p^{2} 4 p^{4} D^{\circ}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | $9.68 \mathrm{E}-01$ |  | $2.93 \mathrm{E}+10$ | $3.00 \mathrm{E}+10$ |  | $1.64 \mathrm{E}-01$ | $1.68 \mathrm{E}-01$ |  | 149.52 |  |
| $2 s 2 p 4 d^{4} p^{\circ}$ | $2 s 2 p 3 p{ }^{4} D^{\text {e }}$ | $2.91 \mathrm{E}-02$ |  | $1.30 \mathrm{E}+09$ | $1.29 \mathrm{E}+09$ |  | $2.84 \mathrm{E}-03$ | $2.81 \mathrm{E}-03$ | $1.43 \mathrm{E}-02^{\text {a }}$ | 155.64 |  |
| $2 s 2 p 4 d^{4} D^{\circ}$ | $2 s 2 p 3 p^{4} D^{\text {e }}$ | $6.59 \mathrm{E}-01$ |  | $1.74 \mathrm{E}+10$ | $1.76 \mathrm{E}+10$ |  | $6.40 \mathrm{E}-02$ | $6.45 \mathrm{E}-02$ |  | 156.44 | $155.15^{\text {a }}$ |
| $2 s 2 p 4 d^{4} P^{\circ}$ | $2 s 2 p 3 p{ }^{4} S^{\text {e }}$ | $9.04 \mathrm{E}-01$ |  | $3.78 \mathrm{E}+10$ | $3.81 \mathrm{E}+10$ | $2.40 \mathrm{E}+10^{\mathrm{a}}$ | $4.31 \mathrm{E}-01$ | $4.35 \mathrm{E}-01$ | $2.67 \mathrm{E}-01^{\text {a }}$ | 159.26 | $157.43^{\text {a }}$ |
| $2 s 2 p 4 d^{4} P^{\circ}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 p^{4} \mathrm{P}^{\mathrm{e}}$ | $8.11 \mathrm{E}-01$ |  | $3.20 \mathrm{E}+10$ | $3.12 \mathrm{E}+10$ | $3.24 \mathrm{E}+10^{\text {a }}$ | $1.26 \mathrm{E}-01$ | $1.23 \mathrm{E}-01$ | $1.25 \mathrm{E}-01^{\text {a }}$ | 162.47 | $160.33^{\text {a }}$ |
| $2 s 2 p 4 d^{4} D^{\circ}$ | $2 s 2 p 3 p{ }^{4} p^{\text {e }}$ | $2.31 \mathrm{E}+00$ |  | $5.36 \mathrm{E}+10$ | $5.35 \mathrm{E}+10$ |  | $3.57 \mathrm{E}-01$ | $3.57 \mathrm{E}-01$ |  | 163.34 |  |
| $2 s 2 p 5 p^{4} D^{\text {e }}$ | $2 p^{2} 3 p^{4} D^{\circ}$ | $1.92 \mathrm{E}-02$ |  | $4.10 \mathrm{E}+08$ | $4.05 \mathrm{E}+08$ |  | $1.73 \mathrm{E}-03$ | $1.71 \mathrm{E}-03$ |  | 167.80 |  |
| $2 s 2 p 5 p^{4} S^{\text {e }}$ | $2 p^{2} 3 p^{4} p^{\circ}$ | $2.81 \mathrm{E}-04$ |  | $2.89 \mathrm{E}+07$ | $3.47 \mathrm{E}+07$ |  | $4.18 \mathrm{E}-05$ | $5.01 \mathrm{E}-05$ |  | 170.12 |  |
| $2 s 2 p 5 p^{4} D^{\text {e }}$ | $2 p^{2} 3 p^{4} p^{\text {o }}$ | $4.45 \mathrm{E}-02$ |  | $9.15 \mathrm{E}+08$ | $8.39 \mathrm{E}+08$ |  | $6.63 \mathrm{E}-03$ | $6.07 \mathrm{E}-03$ |  | 170.19 |  |
| $2 s 2 p 4 s^{4} p^{\circ}$ | $2 s 2 p 3 p{ }^{4} D^{\text {e }}$ | $4.97 \mathrm{E}-01$ |  | $1.82 \mathrm{E}+10$ | $1.92 \mathrm{E}+10$ | $1.55 \mathrm{E}+10^{\text {a }}$ | $4.39 \mathrm{E}-02$ | $5.12 \mathrm{E}-02$ | $4.08 \mathrm{E}-02^{\text {a }}$ | 172.04 |  |
| $2 s 2 p 4 f^{4} D^{\text {e }}$ | $2 s 2 p 3 d^{4} D^{\circ}$ | $1.26 \mathrm{E}+00$ |  | $2.46 \mathrm{E}+10$ | $2.44 \mathrm{E}+10$ |  | $1.11 \mathrm{E}-01$ | $1.10 \mathrm{E}-01$ |  | 173.32 | $172.60^{\text {a }}$ |
| $2 s 2 p 4 f^{4} D^{\text {e }}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 d^{4} \mathrm{P}^{\text {o }}$ | $6.78 \mathrm{E}+00$ |  | $1.26 \mathrm{E}+11$ | $1.25 \mathrm{E}+11$ | $1.03 \mathrm{E}+11^{\text {a }}$ | $9.76 \mathrm{E}-01$ | $9.69 \mathrm{E}-01$ | $7.85 \mathrm{E}-01^{\text {a }}$ | 175.94 |  |
| $2 s 2 p 4 s^{4} p^{\circ}$ | $2 s 2 p 3 p{ }^{4} \mathrm{~S}^{\text {e }}$ | $9.66 \mathrm{E}-02$ |  | $2.97 \mathrm{E}+09$ | $2.99 E+09$ | $2.98 \mathrm{E}+09^{\text {a }}$ | $4.16 \mathrm{E}-02$ | $4.19 \mathrm{E}-02$ | $4.13 \mathrm{E}-02^{\text {a }}$ | 176.48 | $175.24{ }^{\text {a }}$ |
| $2 s 2 p 4 s^{4} p^{\circ}$ | $2 s 2 p 3 p{ }^{4} p^{\text {e }}$ | 5.78E-01 |  | $1.66 \mathrm{E}+10$ | $1.67 \mathrm{E}+10$ | $1.78 \mathrm{E}+10^{\mathrm{a}}$ | $8.11 \mathrm{E}-02$ | $8.14 \mathrm{E}-02$ | $8.51 \mathrm{E}-02$ | 180.42 | $178.84{ }^{\text {a }}$ |
| $2 s 2 p 7 p^{4} S^{\text {e }}$ | $2 s 2 p 4 s{ }^{4} \mathrm{P}^{\circ}$ | $2.64 \mathrm{E}-02$ |  | $2.04 \mathrm{E}+09$ | $2.43 \mathrm{E}+09$ |  | $3.57 \mathrm{E}-03$ | $4.24 \mathrm{E}-03$ |  | 186.98 |  |
| $2 s 2 p 5 d^{4} D^{\circ}$ | $2 p^{2} 3 d^{4} D^{\text {e }}$ | $5.00 \mathrm{E}-03$ |  | $7.61 \mathrm{E}+07$ | $7.65 \mathrm{E}+07$ |  | $4.04 \mathrm{E}-04$ | $4.06 \mathrm{E}-04$ |  | 188.03 |  |
| $2 s 2 p 4 p{ }^{4} p^{\text {e }}$ | $2 s 2 p 3 d^{4} p^{\text {o }}$ | $2.88 \mathrm{E}-02$ |  | $7.20 \mathrm{E}+08$ | $8.05 \mathrm{E}+08$ | $1.68 \mathrm{E}+09^{\text {a }}$ | $3.85 \mathrm{E}-03$ | $4.31 \mathrm{E}-03$ | $8.95 \mathrm{E}-03^{\text {a }}$ | 189.30 | $188.39^{\text {a }}$ |
| $2 s 2 p 4 p{ }^{4} D^{\text {e }}$ | $2 s 2 p 3 d^{4} D^{\circ}$ | 7.63E-02 |  | $1.14 \mathrm{E}+09$ | $1.10 \mathrm{E}+09$ | $1.30 \mathrm{E}+09^{\mathrm{a}}$ | $6.11 \mathrm{E}-03$ | $5.90 \mathrm{E}-03$ | $6.94 \mathrm{E}-03^{\text {a }}$ | 189.49 | $188.82^{\text {a }}$ |
| $2 p^{2} 4 p^{4} D^{\text {o }}$ | $2 p^{2} 3 d^{4} D^{\text {e }}$ | $1.05 \mathrm{E}-01$ |  | $1.52 \mathrm{E}+09$ | $1.68 \mathrm{E}+09$ |  | 8.32E-03 | $9.19 \mathrm{E}-03$ |  | 190.86 |  |
| $2 s 2 p 4 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 3 d^{4} p^{0}$ | $1.54 \mathrm{E}-01$ |  | $1.12 \mathrm{E}+10$ | $1.07 \mathrm{E}+10$ | $5.62 \mathrm{E}+09^{\text {a }}$ | $2.04 \mathrm{E}-02$ | $1.94 \mathrm{E}-02$ | $1.01 \mathrm{E}-02^{\text {a }}$ | 191.02 | $189.80^{\text {a }}$ |
| $2 p^{2} 4 p^{4} S^{\circ}$ | $2 p^{2} 3 d^{4} p^{\text {e }}$ | 6.72E-02 |  | $4.84 \mathrm{E}+09$ | $4.98 \mathrm{E}+09$ |  | $8.88 \mathrm{E}-03$ | $9.14 \mathrm{E}-03$ |  | 191.57 |  |
| $2 s 2 p 4 p^{4} D^{\text {e }}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 d^{4} P^{\circ}$ | 7.73E-02 |  | $1.10 \mathrm{E}+09$ | $9.91 \mathrm{E}+08$ | $6.33 \mathrm{E}+08^{\text {a }}$ | $1.02 \mathrm{E}-02$ | $9.19 \mathrm{E}-03$ | $5.80 \mathrm{E}-03^{\text {a }}$ | 192.63 |  |
| $2 s 2 p 7 p^{4} S^{\text {e }}$ | $2 s 2 p 4 d^{4} P^{\text {o }}$ | $2.34 \mathrm{E}-02$ |  | $1.26 \mathrm{E}+09$ | $1.15 \mathrm{E}+09$ |  | $2.81 \mathrm{E}-03$ | $2.56 \mathrm{E}-03$ |  | 211.16 |  |
| $2 p^{2} 3 d^{4} \mathrm{p}^{\text {e }}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 s^{4} \mathrm{P}^{0}$ | $4.20 \mathrm{E}-03$ |  | $6.44 \mathrm{E}+07$ | $7.33 \mathrm{E}+07$ |  | $4.78 \mathrm{E}-04$ | $5.43 \mathrm{E}-04$ |  | 222.46 |  |
| $2 s 2 p 6 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 4 s{ }^{4}{ }^{\text {o }}$ | $6.58 \mathrm{E}-02$ |  | $2.93 \mathrm{E}+09$ | $2.71 \mathrm{E}+09$ |  | $7.41 \mathrm{E}-03$ | $6.86 \mathrm{E}-03$ |  | 224.94 |  |
| $2 p^{2} 3 d^{4} D^{\text {e }}$ | $2 \mathrm{~s} 2 \mathrm{p} 3 s^{4} \mathrm{P}^{\circ}$ | $1.11 \mathrm{E}-02$ |  | $9.46 \mathrm{E}+07$ | $9.71 \mathrm{E}+07$ | $5.81 \mathrm{E}+07^{\text {a }}$ | $1.23 \mathrm{E}-03$ | $1.26 \mathrm{E}-03$ | $7.65 \mathrm{E}-04^{\text {a }}$ | 228.40 |  |

Table 2 (continued)

| Upper level | Lower level | $S_{P}$ |  | $A_{i k}$ |  |  | $f_{k i}$ |  |  | $\lambda(\AA)$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Present | Others | $A_{l}$ | $A_{v}$ | Others | $f_{l}$ | $f_{v}$ | Others | Present | Others |
| $2 s 2 p 4 d^{4} P^{\text {o }}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | $1.30 \mathrm{E}-03$ |  | $1.67 \mathrm{E}+07$ | $2.55 \mathrm{E}+07$ |  | $1.39 \mathrm{E}-04$ | 2.13E-04 |  | 235.96 |  |
| $2 s 2 p 4 d^{4} D^{\circ}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | $4.98 \mathrm{E}-03$ |  | $3.75 \mathrm{E}+07$ | $4.35 \mathrm{E}+07$ | $3.93 \mathrm{E}+07^{\text {a }}$ | $5.30 \mathrm{E}-04$ | 6.15E-04 | $5.40 \mathrm{E}-04^{\text {a }}$ | 237.80 |  |
| $2 s 2 p 6 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 4 d^{4} P^{\text {o }}$ | 7.27E-02 |  | $2.08 \mathrm{E}+09$ | $1.87 \mathrm{E}+09$ |  | 7.06E-03 | $6.35 \mathrm{E}-03$ |  | 260.88 |  |
| $2 s 2 p 4 f^{4} D^{\text {e }}$ | $2 p^{2} 3 p^{4} D^{\circ}$ | $3.70 \mathrm{E}-02$ |  | $1.94 \mathrm{E}+08$ | $1.93 \mathrm{E}+08$ | $1.94 \mathrm{E}+08^{\text {a }}$ | $2.09 \mathrm{E}-03$ | $2.08 \mathrm{E}-03$ | $2.04 \mathrm{E}-03^{\text {a }}$ | 268.36 | $264.94{ }^{\text {a }}$ |
| $2 s 2 p 4 f^{4} D^{\text {e }}$ | $2 p^{2} 3 p^{4} p^{\circ}$ | $1.02 \mathrm{E}-01$ |  | $5.00 \mathrm{E}+08$ | $5.10 \mathrm{E}+08$ | $3.33 \mathrm{E}+08^{\text {a }}$ | $9.41 \mathrm{E}-03$ | $9.59 \mathrm{E}-03$ | $6.11 \mathrm{E}-03^{\text {a }}$ | 274.53 |  |
| $2 s 2 p 4 s^{4} P^{\circ}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | $3.12 \mathrm{E}-03$ |  | $2.51 \mathrm{E}+07$ | $2.28 \mathrm{E}+07$ | $4.01 \mathrm{E}+07^{\text {a }}$ | $2.86 \mathrm{E}-04$ | $2.60 \mathrm{E}-04$ | $4.51 \mathrm{E}-04^{\text {a }}$ | 275.82 289.80 | $273.58^{\text {a }}$ |
| $2 p^{34} S^{\circ}$ | $2 s 2 p^{2}{ }^{4} p^{\text {e }}$ | 7.57E-01 | $\begin{aligned} & 7.63 \mathrm{E}-01^{\mathrm{b}} \\ & 7.53 \mathrm{E}-01^{\mathrm{c}} \end{aligned}$ | $1.58 \mathrm{E}+10$ | $1.56 \mathrm{E}+10$ |  | 6.61E-02 | 6.57E-02 | $6.65 \mathrm{E}-02^{\mathrm{b}}$ | 289.80 |  |
|  |  |  |  |  |  | $1.58 \mathrm{E}+10^{\mathrm{b}}$ |  |  | $6.60 \mathrm{E}-02^{\mathrm{c}}$ |  | $290.30^{\mathrm{b}}$ |
|  |  |  |  |  |  |  |  |  |  |  | $289.65{ }^{\text {c }}$ |
| $2 \mathrm{~s} 2 \mathrm{p} 4 p^{4} P^{\mathrm{e}}$ | $2 p^{2} 3 p^{4} p^{\text {o }}$ | $1.33 \mathrm{E}-02$ |  | $7.65 \mathrm{E}+07$ | $8.64 \mathrm{E}+07$ | $1.11 \mathrm{E}+08^{\text {a }}$ | $1.09 \mathrm{E}-03$ | $1.23 \mathrm{E}-03$ | $1.56 \mathrm{E}-03^{\text {a }}$ | 308.50 | $305.47{ }^{\text {a }}$ |
| $2 s 2 p 4 p{ }^{4}$ e | $2 p^{2} 3 p^{4} D^{\text {o }}$ | $1.73 \mathrm{E}-03$ |  | $5.93 \mathrm{E}+06$ | $4.24 \mathrm{E}+06$ |  | $8.50 \mathrm{E}-05$ | $6.08 \mathrm{E}-05$ |  | 309.22 | $305.16^{\text {a }}$ |
| $2 \mathrm{~s} 2 p 4 p{ }^{4} S^{\text {e }}$ | $2 p^{2} 3 p^{4} p^{\circ}$ | $6.35 \mathrm{E}-03$ |  | $1.05 \mathrm{E}+08$ | $9.64 \mathrm{E}+07$ | $9.63 \mathrm{E}+07^{\text {a }}$ | 5.13E-04 | $4.72 \mathrm{E}-04$ | $4.57 \mathrm{E}-04{ }^{\text {a }}$ | 313.08 | $309.21^{\text {a }}$ |
| $2 s 2 p 4 p{ }^{4} D^{\text {e }}$ | $2 p^{2} 3 p^{4} p^{\text {o }}$ | $1.06 \mathrm{E}-02$ |  | $3.37 \mathrm{E}+07$ | $3.60 \mathrm{E}+07$ | $7.19 \mathrm{E}+07^{\text {a }}$ | 8.48E-04 | $9.06 \mathrm{E}-04$ | $1.77 \mathrm{E}-03^{\text {a }}$ | 317.45 |  |
| $2 s 2 p 5 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 4 s{ }^{4} P^{\text {o }}$ | $4.66 \mathrm{E}-01$ |  | $6.40 \mathrm{E}+09$ | $5.13 \mathrm{E}+08$ |  | 3.54E-02 | $2.84 \mathrm{E}-02$ |  | 332.91 |  |  |
| $2 s 2 p 5 p{ }^{4} D^{\text {e }}$ | $2 s 2 p 4 s{ }^{4} P^{\text {o }}$ | $2.64 \mathrm{E}+00$ |  | $7.23 \mathrm{E}+09$ | $6.82 \mathrm{E}+09$ |  | $2.01 \mathrm{E}-01$ | $1.89 \mathrm{E}-01$ |  |  |  |  |
| $2 s 2 p 4 p{ }^{4} P^{\text {e }}$ | $2 p^{2} 3 p^{4} S^{\circ}$ | $2.35 \mathrm{E}-03$ |  | $1.04 \mathrm{E}+07$ | $7.52 \mathrm{E}+06$ | $1.10 \mathrm{E}+07^{\text {a }}$ | $5.29 \mathrm{E}-04$ | $3.83 \mathrm{E}-04$ | $5.81 \mathrm{E}-04^{\text {a }}$ | 336.36340.36 | $342.66^{\text {a }}$ |
| $2 p^{2} 3 p^{4} p^{\text {o }}$ | $2 s 2 p 3 p{ }^{4} D^{\text {e }}$ | $7.46 \mathrm{E}-01$ |  | $3.20 \mathrm{E}+09$ | $2.95 E+09$ | $3.07 \mathrm{E}+09^{\text {a }}$ | $3.33 \mathrm{E}-02$ | $3.07 \mathrm{E}-02$ | $3.15 \mathrm{E}-02^{\text {a }}$ |  |  |
| $2 p^{2} 3 p^{4} S^{\circ}$ | $2 s 2 p 3 p{ }^{4} P^{\text {e }}$ | $5.09 \mathrm{E}-01$ |  | $6.53 \mathrm{E}+09$ | $6.37 \mathrm{E}+09$ | $8.25 \mathrm{E}+09^{\text {a }}$ | $3.79 \mathrm{E}-02$ | $3.69 \mathrm{E}-02$ | $4.38 \mathrm{E}-02^{\text {a }}$ | 340.53 | $326.60^{\text {a }}$ |
| $2 p^{2} 3 s^{4} P^{\text {e }}$ | $2 s 2 p 3 s{ }^{4} P^{\text {o }}$ | $1.40 \mathrm{E}+00$ |  | $5.92 \mathrm{E}+09$ | $5.79 \mathrm{E}+09$ | $6.25 \mathrm{E}+09^{\text {a }}$ | $1.04 \mathrm{E}-01$ | $1.01 \mathrm{E}-01$ | $1.08 \mathrm{E}-01^{\text {a }}$ | 341.32 | $339.64{ }^{\text {a }}$ |
| $2 s 2 p 5 d^{4} D^{\circ}$ | $2 s 2 p 4 p{ }^{4}{ }^{\text {e }}$ | $5.57 \mathrm{E}-01$ |  | $1.33 \mathrm{E}+09$ | $1.39 \mathrm{E}+09$ |  | $2.43 \mathrm{E}-02$ | $2.54 \mathrm{E}-02$ |  | 348.97 |  |
| $2 p^{2} 3 d^{4} p^{\text {e }}$ | $2 s 2 p 3 d{ }^{4} P^{\text {o }}$ | $4.25 \mathrm{E}-01$ |  | $1.67 \mathrm{E}+09$ | $1.68 \mathrm{E}+09$ |  | $3.07 \mathrm{E}-02$ | $3.09 \mathrm{E}-02$ |  | 350.20 |  |
| $2 p^{2} 3 p^{4} D^{\text {o }}$ | $2 s 2 p 3 p{ }^{4} D^{\text {e }}$ | $1.98 \mathrm{E}+00$ |  | $4.69 \mathrm{E}+09$ | $4.55 \mathrm{E}+09$ | $4.62 \mathrm{E}+09^{\text {a }}$ | $8.60 \mathrm{E}-02$ | 8.36E-02 | $8.12 \mathrm{E}-02^{\text {a }}$ | 350.35 | $349.02^{\text {a }}$ |
| $2 p^{2} 3 d^{4} D^{\text {e }}$ | $2 s 2 p 3 d^{4} D^{\text {o }}$ | 8.82E-02 |  | $2.01 \mathrm{E}+08$ | $1.88 \mathrm{E}+08$ | $2.78 \mathrm{E}+08^{\text {a }}$ | $3.78 \mathrm{E}-03$ | 3.54E-03 | $5.45 \mathrm{E}-03^{\text {a }}$ | 354.03 | $362.76{ }^{\text {a }}$ |
| $2 p^{2} 4 p^{4} S^{\circ}$ | $2 s 2 p 4 p{ }^{4} P^{\text {e }}$ | $4.91 \mathrm{E}-01$ |  | $5.41 \mathrm{E}+09$ | $5.18 \mathrm{E}+09$ |  | $3.47 \mathrm{E}-02$ | $3.47 \mathrm{E}-02$ |  | 358.03 |  |
| $2 p^{2} 3 p^{4} p^{\circ}$ | $2 s 2 p 3 p{ }^{4} S^{\text {e }}$ | 4.89E-01 |  | $1.80 \mathrm{E}+09$ | $1.77 \mathrm{E}+09$ | $1.84 \mathrm{E}+09^{\text {a }}$ | $1.04 \mathrm{E}-01$ | $1.02 \mathrm{E}-01$ | $1.04 \mathrm{E}-01^{\text {a }}$ | 358.16 | $354.44^{\text {a }}$ |
| $2 s 2 p 4 d{ }^{4} P^{\text {o }}$ | $2 p^{2} 3 d^{4} D^{\text {e }}$ | $3.04 \mathrm{E}-02$ |  | $1.11 \mathrm{E}+08$ | $1.22 \mathrm{E}+08$ | $2.10 \mathrm{E}+08^{\text {a }}$ | $1.29 \mathrm{E}-03$ | $1.41 \mathrm{E}-03$ | $2.26 \mathrm{E}-03^{\text {a }}$ | 358.51 |  |
| $2 p^{2} 4 p{ }^{4} D^{\circ}$ | $2 s 2 p 4 p{ }^{4} D^{\text {e }}$ | $1.97 \mathrm{E}+00$ |  | $4.31 \mathrm{E}+09$ | $4.01 \mathrm{E}+09$ |  | 8.32E-02 | 7.73E-02 |  | 358.83362.76 |  |
| $2 s 2 p 4 d^{4} D^{\circ}$ | $2 p^{2} 3 d^{4} D^{\text {e }}$ | $6.47 \mathrm{E}-03$ |  | $1.37 \mathrm{E}+07$ | $1.48 \mathrm{E}+07$ |  | $2.71 \mathrm{E}-04$ | 2.92E-04 |  |  |  |
| $2 p^{2} 3 d^{4} D^{\text {e }}$ | $2 s 2 p 3 d{ }^{4} P^{\text {o }}$ | $4.07 \mathrm{E}-01$ |  | $8.47 \mathrm{E}+08$ | $7.87 \mathrm{E}+08$ | $7.05 \mathrm{E}+08^{\text {a }}$ | $2.82 \mathrm{E}-02$ | $2.62 \mathrm{E}-02$ | $2.45 \mathrm{E}-02^{\text {a }}$ | 365.15 |  |
| $2 s 2 p 4 d^{4} P^{\text {o }}$ | $2 p^{2} 3 d^{4} p^{\text {e }}$ | $9.46 \mathrm{E}-03$ |  | $3.05 \mathrm{E}+07$ | $2.98 \mathrm{E}+07$ |  | $6.40 \mathrm{E}-04$ | $6.25 \mathrm{E}-04$ |  | 374.19 |  |
| $2 p^{2} 3 p^{4} p^{\circ}$ | $2 s 2 p 3 p{ }^{4} p^{\text {e }}$ | 5.03E-01 |  | $1.61 \mathrm{E}+09$ | $1.48 \mathrm{E}+09$ | $1.73 \mathrm{E}+09^{\text {a }}$ | $3.40 \mathrm{E}-02$ | $3.11 \mathrm{E}-02$ | $3.55 \mathrm{E}-02^{\text {a }}$ | 374.80 | $369.46^{\text {a }}$ |
| $2 p^{2} 3 p^{4} D^{\text {o }}$ | $2 s 2 p 3 p{ }^{4} P^{\text {e }}$ | $7.38 \mathrm{E}-01$ |  | $1.29 \mathrm{E}+09$ | $1.23 \mathrm{E}+09$ | $1.31 \mathrm{E}+09^{\text {a }}$ | $4.83 \mathrm{E}-02$ | $4.61 \mathrm{E}-02$ | $4.80 \mathrm{E}-02^{\text {a }}$ | 386.94 |  |
| $2 s 2 p 5 p^{4} D^{\text {e }}$ | $2 s 2 p 4 d{ }^{4} D^{\text {o }}$ | $5.11 \mathrm{E}-01$ |  | $7.36 \mathrm{E}+08$ | $6.22 \mathrm{E}+08$ |  | $1.88 \mathrm{E}-02$ | $1.59 \mathrm{E}-02$ |  | 412.93 |  |
| $2 s 2 p 5 p^{4} S^{\text {e }}$ | $2 s 2 p 4 d^{4} P^{\text {o }}$ | $4.66 \mathrm{E}-01$ |  | $3.23 \mathrm{E}+09$ | $3.75 \mathrm{E}+09$ |  | 2.82E-02 | $3.28 \mathrm{E}-02$ |  | 418.17 |  |
| $2 s 2 p 5 p{ }^{4} D^{\text {e }}$ | $2 s 2 p 4 d^{4} P^{\text {o }}$ | $1.41 \mathrm{E}-03$ |  | $1.94 \mathrm{E}+06$ | $1.86 \mathrm{E}+06$ |  | $8.51 \mathrm{E}-05$ | $8.14 \mathrm{E}-05$ |  | 418.59 |  |
| $2 s 2 p 5 d^{4} D^{\text {o }}$ | $2 s 2 p 4 f^{4} D^{\text {e }}$ | 4.96E-02 |  | $6.71 \mathrm{E}+07$ | $7.00 \mathrm{E}+07$ |  | $1.79 \mathrm{E}-03$ | $1.86 \mathrm{E}-03$ |  | 421.38 |  |
| $2 p^{2} 4 p^{4} D^{\circ}$ | $2 s 2 p 4 f^{4} D^{\text {e }}$ | $2.28 \mathrm{E}-03$ |  | $2.80 \mathrm{E}+06$ | $2.00 \mathrm{E}+06$ |  | $7.96 \mathrm{E}-05$ | $5.71 \mathrm{E}-05$ |  | 435.84 |  |
| $2 s 2 p 4 s^{4} P^{\circ}$ | $2 p^{2} 3 d^{4} D^{\text {e }}$ | $1.25 \mathrm{E}-02$ |  | $2.17 \mathrm{E}+07$ | $2.09 \mathrm{E}+07$ | $2.56 \mathrm{E}+07^{\text {a }}$ | $4.13 \mathrm{E}-04$ | $4.04 \mathrm{E}-04$ | $4.59 \mathrm{E}-04^{\text {a }}$ | 459.37 |  |
| $2 s 2 p 4 s^{4} P^{\circ}$ | $2 p^{2} 3 d^{4} p^{\text {e }}$ | $4.44 \mathrm{E}-03$ |  | $6.55 E+06$ | $7.39 \mathrm{E}+06$ | $1.40 \mathrm{E}+07^{\text {a }}$ | $2.31 \mathrm{E}-04$ | $2.61 \mathrm{E}-04$ | $5.08 \mathrm{E}-04^{\text {a }}$ | 485.44 | $493.08^{\text {a }}$ |
| $2 p^{2} 3 s^{4} P^{\text {e }}$ | $2 s 2 p 3 d^{4} D^{\text {o }}$ | $5.04 \mathrm{E}-02$ |  | $2.22 \mathrm{E}+07$ | $3.84 \mathrm{E}+07$ | $2.42 \mathrm{E}+07^{\text {a }}$ | $1.05 \mathrm{E}-03$ | $1.82 \mathrm{E}-03$ | $1.21 \mathrm{E}-03^{\text {a }}$ | 726.71 |  |
| $2 p^{2} 3 s^{4} p^{\text {e }}$ | $2 s 2 p 3 d^{4} P^{\circ}$ | $1.62 \mathrm{E}-02$ |  | $5.89 \mathrm{E}+06$ | $8.26 \mathrm{E}+06$ | $5.77 \mathrm{E}+06^{\text {a }}$ | $5.30 \mathrm{E}-04$ | $7.44 \mathrm{E}-04$ | $5.34 \mathrm{E}-04^{\text {a }}$ | 775.19 |  |
| $2 p^{2} 3 p^{4} S^{\circ}$ | $2 p^{2} 3 s^{4} P^{\text {e }}$ | $1.98 \mathrm{E}+00$ |  | $1.06 \mathrm{E}+09$ | $1.03 \mathrm{E}+09$ | $1.41 \mathrm{E}+09^{\text {a }}$ | $5.10 \mathrm{E}-02$ | 4.95E-02 | $5.57 \mathrm{E}-02^{\text {a }}$ | 980.83 | $888.85^{\mathrm{a}}$ |
| $2 s 2 p 3 p{ }^{4} p^{\text {e }}$ | $2 s 2 p 3 s{ }^{4} P^{\text {o }}$ | $5.12 \mathrm{E}+00$ |  | $8.97 \mathrm{E}+08$ | $9.05 \mathrm{E}+08$ | $8.63 \mathrm{E}+08^{\text {a }}$ | $1.31 \mathrm{E}-01$ | $1.32 \mathrm{E}-01$ | $1.28 \mathrm{E}-01^{\text {a }}$ | 987.41 | $992.26^{\text {a }}$ |
| $2 s 2 p 3 d{ }^{4}{ }^{\text {o }}$ | $2 s 2 p 3 p{ }^{4} D^{\text {e }}$ | $5.32 \mathrm{E}-01$ |  | $6.48 \mathrm{E}+07$ | $5.67 \mathrm{E}+07$ |  | $7.25 \mathrm{E}-03$ | $6.34 \mathrm{E}-03$ |  | 1114.83 |  |
| $2 s 2 p 3 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 3 s{ }^{4} p^{\circ}$ | $1.59 \mathrm{E}+00$ |  | $5.66 \mathrm{E}+08$ | $5.59 \mathrm{E}+08$ | $5.21 \mathrm{E}+08^{\text {a }}$ | 3.58E-02 | 3.54E-02 | $3.30 \mathrm{E}-02^{\text {a }}$ | 1125.13 | $1120.61^{\text {a }}$ |
| $2 p^{2} 3 d^{4} p^{\text {e }}$ | $2 p^{2} 3 p^{4} p^{\circ}$ | $2.56 \mathrm{E}+00$ |  | $2.34 \mathrm{E}+08$ | $2.07 \mathrm{E}+08$ |  | 5.28E-02 | $4.68 \mathrm{E}-02$ |  | 1227.88 |  |
| $2 s 2 p 3 d^{4} D^{\circ}$ | $2 s 2 p 3 p{ }^{4} D^{\text {e }}$ | $1.49 \mathrm{E}+00$ |  | $8.05 E+07$ | $7.53 \mathrm{E}+07$ |  | $1.84 \mathrm{E}-02$ | $1.72 \mathrm{E}-02$ |  | 1233.13 |  |
| $2 p^{2} 3 d^{4} D^{\text {e }}$ | $2 p^{2} 3 p^{4} D^{\text {o }}$ | $2.25 \mathrm{E}+00$ |  | $1.08 \mathrm{E}+08$ | $9.32 \mathrm{E}+07$ | $8.28 \mathrm{E}+07^{\text {a }}$ | $2.66 \mathrm{E}-02$ | $2.29 \mathrm{E}-02$ | $2.29 \mathrm{E}-02^{\text {a }}$ | 1279.97 | $1356.40^{\text {a }}$ |
| $2 s 2 p 4 f^{4} D^{\text {e }}$ | $2 s 2 p 4 s^{4} P^{\text {o }}$ | 8.07E-02 |  | $3.71 \mathrm{E}+06$ | $4.35 \mathrm{E}+06$ |  | $1.57 \mathrm{E}-03$ | $1.84 \mathrm{E}-03$ |  | 1301.87 |  |
| $2 p^{2} 3 p^{4} p^{\circ}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | $5.49 \mathrm{E}+00$ |  | $3.93 \mathrm{E}+08$ | $3.75 \mathrm{E}+08$ | $4.16 \mathrm{E}+08^{\text {a }}$ | $1.04 \mathrm{E}-01$ | $9.95 \mathrm{E}-02$ | $1.06 \mathrm{E}-01^{\text {a }}$ | 1331.39 | $1298.51{ }^{\text {a }}$ |
| $2 s 2 p 3 d^{4} P^{\text {o }}$ | $2 s 2 p 3 p{ }^{4} S^{\text {e }}$ | $2.80 \mathrm{E}+00$ |  | $2.00 \mathrm{E}+08$ | $1.96 \mathrm{E}+08$ | $1.80 \mathrm{E}+08^{\text {a }}$ | $1.60 \mathrm{E}-01$ | $1.56 \mathrm{E}-01$ | $1.31 \mathrm{E}-01^{\text {a }}$ | 1331.59 | $1271.02^{\text {a }}$ |
| $2 s 2 p 3 p^{4} D^{\text {e }}$ | $2 s 2 p 3 s^{4} P^{\text {o }}$ | $8.09 \mathrm{E}+00$ |  | $3.36 E+08$ | $4.53 \mathrm{E}+08$ |  | $1.52 \mathrm{E}-01$ | $2.05 \mathrm{E}-01$ |  | 1346.30 |  |
| $2 p^{2} 3 d^{4} D^{\text {e }}$ | $2 p^{2} 3 p 4 P$ o | $6.34 \mathrm{E}+00$ |  | $2.18 \mathrm{E}+08$ | $1.57 \mathrm{E}+08$ | $1.56 \mathrm{E}+08^{\text {a }}$ | $1.12 \mathrm{E}-01$ | 8.08E-02 | $9.35 \mathrm{E}-02^{\text {a }}$ | 1433.69 |  |
| $2 p^{2} 3 p^{4} D^{\text {o }}$ | $2 p^{2} 3 s^{4} p^{\text {e }}$ | $9.16 E+00$ |  | $2.76 \mathrm{E}+08$ | $2.66 \mathrm{E}+08$ | $2.91 \mathrm{E}+08^{\text {a }}$ | $1.55 \mathrm{E}-01$ | $1.49 \mathrm{E}-01$ | $1.58 \mathrm{E}-01^{\text {a }}$ | 1498.53 |  |
| $2 s 2 p 3 d^{4} P^{\text {o }}$ | $2 s 2 p 3 p{ }^{4} p^{\text {e }}$ | $1.31 \mathrm{E}+00$ |  | $5.47 \mathrm{E}+07$ | $5.57 \mathrm{E}+07$ |  | $2.09 \mathrm{E}-02$ | $2.12 \mathrm{E}-02$ |  | 1594.83 |  |
| $2 p^{2} 3 d^{4} p^{\text {e }}$ | $2 p^{2} 3 p^{4} S^{\circ}$ | $2.32 \mathrm{E}+00$ |  | $6.38 \mathrm{E}+07$ | $6.62 \mathrm{E}+07$ |  | $9.63 \mathrm{E}-02$ | $9.99 \mathrm{E}-02$ |  | 1831.65 |  |
| $2 s 2 p 3 d^{4} D^{\text {o }}$ | $2 s 2 p 3 p{ }^{4} P^{\text {e }}$ | $5.07 \mathrm{E}+00$ |  | $8.18 \mathrm{E}+07$ | $8.09 \mathrm{E}+07$ |  | 6.94E-02 | $6.86 \mathrm{E}-02$ |  | 1848.52 |  |
| $2 s 2 p 4 p{ }^{4} p^{\text {e }}$ | $2 s 2 p 4 s^{4} P^{\circ}$ | $1.97 \mathrm{E}+01$ |  | $1.64 \mathrm{E}+08$ | $1.69 \mathrm{E}+08$ | $1.77 \mathrm{E}+08^{\text {a }}$ | $1.83 \mathrm{E}-01$ | $1.88 \mathrm{E}-01$ | $1.85 \mathrm{E}-01^{\text {a }}$ | 2724.32 | $2576.65^{\text {a }}$ |
| $2 s 2 p 4 d^{4} p^{\circ}$ | $2 s 2 p 4 p{ }^{4} D^{\text {e }}$ | $5.66 \mathrm{E}-01$ |  | $3.65 \mathrm{E}+06$ | $5.60 \mathrm{E}+06$ |  | $2.90 \mathrm{E}-03$ | $4.44 \mathrm{E}-03$ |  | 2969.71 |  |
| $2 s 2 p 4 p{ }^{4} S^{\text {e }}$ | $2 s 2 p 4 s^{4} P^{\circ}$ | $5.40 \mathrm{E}+00$ |  | $8.94 \mathrm{E}+07$ | $9.27 \mathrm{E}+07$ | $7.49 \mathrm{E}+07^{\text {a }}$ | $4.37 \mathrm{E}-02$ | $4.53 \mathrm{E}-02$ | $2.74 \mathrm{E}-02^{\text {a }}$ | 3128.34 | $2879.18^{\text {a }}$ |
| $2 s 2 p 4 d^{4} D^{\circ}$ | $2 s 2 p 4 p{ }^{4} D^{\text {e }}$ | $7.97 \mathrm{E}+00$ |  | $2.27 \mathrm{E}+07$ | $2.26 \mathrm{E}+07$ |  | 3.68E-02 | $3.66 \mathrm{E}-02$ |  | 3289.36 | $3303.26^{\text {a }}$ |
| $2 s 2 p 4 d^{4} p^{\circ}$ | $2 s 2 p 4 p{ }^{4}$ S | $1.09 \mathrm{E}+01$ |  | $4.62 \mathrm{E}+07$ | $5.33 \mathrm{E}+07$ | $2.60 \mathrm{E}+07^{\text {a }}$ | $2.42 \mathrm{E}-01$ | $2.79 \mathrm{E}-01$ | $1.34 \mathrm{E}-01^{\text {a }}$ | 3415.35 | $3374.80^{\text {a }}$ |
| $2 s 2 p 4 p{ }^{4} D^{\text {e }}$ | $2 s 2 p 4 s^{4} P^{\circ}$ | $3.01 \mathrm{E}+01$ |  | $6.38 E+07$ | $6.77 \mathrm{E}+07$ | $8.37 \mathrm{E}+07^{\text {a }}$ | $2.10 \mathrm{E}-01$ | $2.23 \mathrm{E}-01$ | $2.30 \mathrm{E}-01{ }^{\text {a }}$ | 3626.86 |  |
| $2 s 2 p 4 d^{4} P^{\circ}$ | $2 s 2 p 4 p{ }^{4} P^{\text {e }}$ | $8.43 \mathrm{E}+00$ |  | $2.10 \mathrm{E}+07$ | $1.66 \mathrm{E}+07$ |  | $5.23 \mathrm{E}-02$ | 4.15E-02 |  | 4075.15 |  |
| $2 s 2 p 4 f^{4} D^{\text {e }}$ | $2 s 2 p 4 d^{4} D^{\text {o }}$ | $3.45 \mathrm{E}+00$ |  | $2.34 \mathrm{E}+06$ | $1.97 \mathrm{E}+06$ |  | $9.86 \mathrm{E}-03$ | $8.31 \mathrm{E}-03$ |  | 5308.04 |  |
| $2 s 2 p 4 f^{4} D^{\text {e }}$ | $2 s 2 p 4 d^{4} P^{\text {o }}$ | $1.87 \mathrm{E}+01$ |  | $7.14 \mathrm{E}+06$ | $6.06 \mathrm{E}+06$ |  | 7.36E-02 | $6.25 \mathrm{E}-02$ |  | 6423.83 |  |
| $2 s 2 p 5 d^{4} D^{\circ}$ | $2 s 2 p 5 p{ }^{4}{ }^{\text {e }}$ | $1.93 \mathrm{E}+01$ |  | $5.35 \mathrm{E}+06$ | $5.62 \mathrm{E}+06$ |  | $4.10 \mathrm{E}-02$ | $4.31 \mathrm{E}-02$ |  | 7151.82 |  |

a. CHIANT-Ref. [27]. b. NIST-Ref. [26]. c. MCHF-Ref. [21].

The data in bold font in the last column indicates experimental data from CHIANTI [27].


Fig. 2. Comparisons of the length form of oscillator strengths with the velocity form of oscillator strengths calculated in this work.
states especially for the high-lying states with the relative energy $E_{S}>21 \mathrm{Ry}$. However, no more experimental data are available for these states, we could not give a comparison. More accurate experimental data are needed to identify it.

Using the calculated wave functions and energy levels, we further systematically study the electric-dipole transitions between these quartet ${ }^{4} L^{\mathrm{e}, \mathrm{o}}(L=S, P, D)$ Rydberg series in $\mathrm{Si}^{9+}$ ion. In this work, the oscillator strengths and transition probabilities are calculated in the frame of non-relativistic wave functions. In order to eliminate the largest part of uncertainty related to inaccuracy of the non-relativistic energy levels, the oscillator strengths and transition probabilities are rescaled to wavelengths from the relativistic levels. In Table 2, we list the calculated line strengths $S_{P}$, oscillator strengths $f_{k i}$, transition probabilities $A_{i k}$, and transition wavelengths for the electric-dipole transitions between these ${ }^{4} L^{\text {e,o }}$ Rydberg series terms in $\mathrm{Si}^{9+}$ ion, except the weak transitions with line strength $S_{P}<3 \times 10^{-5}$ (a.u.) All the oscillator strengths and transition probabilities are given in two forms: the length and velocity gauges. The consistency between the results from these two gauges is usually used as an indicator of the overall accuracy the calculated wave functions. In order to estimate the uncertainty of the calculated oscillator strengths, in Fig. 2, we compare the calculated oscillator strengths in the form of $f_{l}$ with the velocity form $f_{v}$, along with line strengths $S_{P}$ in $\mathrm{Si}^{9+}$ ion. It can be deduced from Fig. 2 that the deviations between the results in two gauges in 100 transition are less than $10 \%$, in 20 transition are less than $20 \%$, and in 12 transitions are more than $20 \%$. Fig. 3 presents the comparison of the calculated oscillator strengths $f_{l}$ with the referenced oscillator strength values from CHIANTI [27] database calculated with the AS method. Fig. 3 shows that large discrepancies exist between our calculated oscillator strengths and those from the AS theoretical method, especially in the


Fig. 3. Comparisons of the length form of oscillator strengths $f_{l}$ with the referenced data from CHIANTI [27] database.


Fig. 4. Comparisons of the calculated transition wavelengths with the referenced data from CHIANTI [27] database.
range of $S_{P}<0.1$ (a.u.). Unfortunately, limited theoretical or experimental data are available, so we could not make further overall comparison. Table 2 also lists the transition parameters from the NIST [26] and MCHF theoretical data [21] for three transitions: $2 p^{3} S^{4}-2 s 2 p^{2}$ ${ }^{4} P^{\mathrm{e}}, 2 s 2 p 3 s{ }^{4} P^{\mathrm{o}}-2 s 2 p^{2}{ }^{4} P^{\mathrm{e}}$, and $2 s 2 p 3 d{ }^{4} D^{\mathrm{o}}-2 s 2 p^{2}{ }^{4} P^{\mathrm{e}}$. Good agreements are obtained when comparing the calculated results in this work and the referenced data for these three transitions. For example, for the $2 p^{3}{ }^{4} S^{\mathrm{o}}-2 s 2 p^{2}{ }^{4} P^{\mathrm{e}}$ transition, our calculated line strength $S_{P}=0.757$ and oscillator strength $f_{l}=0.0662$, agree well with the NIST experimental results $S=0.763, f=0.0665$, and are also in accord with the MCHF theoretical data $S=0.753, f=0.0660$. As the transition probabilities are proportional to oscillator strengths, the deviations $A_{l}$ from $A_{v}$ values should have the similar magnitude as the deviations of oscillator strengths discussed above.

In the last two columns of Table 2, we tabulate the calculated transition wavelengths, together with the NIST experimental results [26] wherever available, the data from CHIANTI database [27], and the theoretical data from the MCHF calculations [21]. Our calculated data are in good agreement with the NIST experimental results and the MCHF theoretical data. For example, the present calculated wavelength value $54.58 \AA$ for the transition of $2 s 2 p 3 s{ }^{4} P^{\circ}-2 s 2 p^{2}{ }^{4} P^{\mathrm{e}}$, agree well with the NIST experimental value $54.582 \AA$ [26] and MCHF theoretical data $54.56 \AA$ [21]. Fig. 4 gives the comparisons of the present transition wavelengths with the data from the CHIANTI database [27]. Here, 4 transition wavelengths from CHIANTI are experimental values, which are labelled with bold font in the last column of Table 2. The remaining referenced wavelengths from CHIANTI are calculated by the AS method [25]. As can be seen from Fig. 4, our calculated wavelengths generally agree well with the CHIANTI data in the range of $\lambda \leq 500 \AA$, but do not fit well in the range of $\lambda>500 \AA$. It is because the wavelength of long wave is more sensitive to the deviation of energy levels. The largest deviation is for the transition $2 p^{2} 3 p{ }^{4} S^{\circ}-2 p^{2} 3 s{ }^{4} P^{\mathrm{e}}$, our calculated wavelength $980.83 \AA$ is about $9.4 \%$ larger than the AS theoretical value $888.85 \AA$ [25]. The large deviation is resulting from the disagreement in the energy level calculation of the $2 p^{2} 3 p{ }^{4} S^{\circ}$ state. Most of these transition wavelengths are in the range of EUV or soft X-ray, which are particular importance in astrophysical applications. The present calculated data would provide theoretical support for identification of astrophysical spectral lines and other applications.

## 4. Conclusions

The radiative transition processes of the quartet ${ }^{4} L^{\mathrm{e}, \mathrm{o}}(L=S, P, D)$ Rydberg series in boron-like $\mathrm{Si}^{9+}$ ion are systematically studied, by employing the multi-configuration Rayleigh-Ritz variation method. The
energy levels and the radiative transition parameters are calculated in detail. The calculated results have been compared with available accurate experimental data from NIST and CHIANTI database and other theoretical data. Good agreements are found between the data obtained in this work and the experimental results. The good agreements between the oscillator strengths from the length and velocity gauges indicates that the calculated values should be generally accurate. However, there are still large discrepancies for some high-lying Rydberg states, for which further studies are needed. The present results will provide reliable theoretical data for further experiments.

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